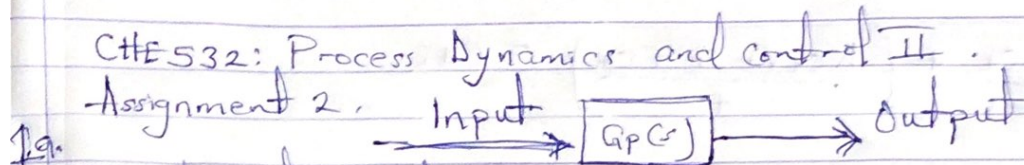


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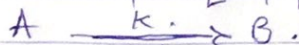
Matric No: 14/EATG01/009.

CHE532: Process Dynamics and control II.

Assignment 2.



Isothermal liquid phase process.



Input Variable =  $C_{Ai}$  (mol/dm<sup>3</sup>).

Output Variable =  $C_A$  (mol/dm<sup>3</sup>).

First Order reaction

$$-r_A = k C_A$$

$$r_A = -k C_A \quad \xrightarrow{F_A V} \text{Generation.}$$

Taking a general material balance.

Input + Generation - Output - Consumption = Accumulation.

$$\dot{m}_{in} + (r_A V) - \dot{m}_{out} = \frac{dm}{dt}$$

$$\dot{m} = \frac{\text{mol}}{\text{dm}^3} \times \frac{\text{dm}^3}{\text{min}}$$

$$\dot{m} = F \cdot C_A$$

$$m (\text{mol}) = \frac{\text{mol}}{\text{dm}^3} \times \text{dm}^3$$

$$m = C_A V$$

$$\frac{dm (\text{mol})}{dt (\text{min})} = \frac{\text{mol}}{\text{min}}$$

$$F C_{Ai} - F C_A + r_A V_A = \frac{dm}{dt}$$

$$F C_{Ai} - F C_A - k C_A V_A = \frac{dm}{dt}$$

$$F C_{Ai} - F C_A - k C_A V_A = \frac{d(C_A \cdot V_A)}{dt}$$

Assuming constant volume

$$F C_{Ai} - F C_A - k C_A V_A = V_A \frac{dC_A}{dt}$$

$$V_A \frac{dC_A}{dt} = F C_{Ai} - F C_A - k C_A V_A$$

$$\frac{V_A}{F} \frac{dC_A}{dt} = C_{Ai} - C_A - \frac{k C_A V_A}{F}$$

$$\frac{V_A}{F} \frac{dC_A}{dt} = C_{Ai} - \frac{k C_A V_A}{F} - \frac{C_A}{1}$$

$$\frac{V_A}{F} \frac{dC_A}{dt} = C_{Ai} + \left( \frac{-k C_A V_A - F C_A}{F} \right)$$



$$\frac{V_A}{F} \frac{dC_A}{dt} + \left( \frac{K_C V_A + F C_A}{F} \right) = C_A$$

$$\frac{V_A}{F} \frac{dC_A}{dt} + \frac{F C_A + K_C V_A}{F} = C_A$$

$$\frac{V_A}{F} \frac{dC_A}{dt} + C_A \left( \frac{F + K_C V_A}{F} \right) = C_A$$

dividing through by  $\frac{F + K_C V_A}{F}$

$$\frac{V_A}{F} \frac{dC_A}{dt} \times \frac{F}{F + K_C V_A} + C_A = C_A \times \frac{F}{F + K_C V_A}$$

$$\frac{V_A}{F + K_C V_A} \frac{dC_A}{dt} + C_A = \frac{F}{F + K_C V_A} C_A$$

$$T_p \frac{dy}{dt} + y = K_p u$$

$$G_p(s) = \frac{K_p}{T_p s + 1}$$

$$G_p(s) = \left( \frac{F}{F + K_C V_A} \right) / \left( \frac{V_A}{F + K_C V_A} s + 1 \right)$$

$$K_p = \frac{F}{F + K_C V_A}, \quad T_p = \frac{V_A}{F + K_C V_A}$$

$$K_p = \frac{0.091}{0.091 + 0.03 \times 2.5} = 0.55$$

b) Given that:

$$V = 2.5 \text{ m}^3$$

$$F = 0.091 \text{ m}^3/\text{min}$$

$$K = 0.03 \text{ min}^{-1}$$

$$T_p = \frac{2.5}{0.091 + 0.03 \times 2.5} = 15.06$$

with the aid of a matlab/simulink, run via m-file.  
(i) The graphical open-loop dynamic response of the process to a 2 unit step change in the input variable applied at  $t = 0.2 \text{ min}$ .

$$G_p(s) = \frac{K_p}{T_p s + 1} = \frac{0.55}{15.06 s + 1}$$





Transfer

from the analysis,

The steady state value  $= K_p \Delta U = 0.55 \times 2 = 1.1$  corresponding to the graph as shown in the m-file.

Time at which it gets to steady state  $= 5\tau_p = 5 \times 15.06 = 75.3s$

- ii Closed loop dynamic response of the process.  
 3 unit step change at  $t = 0.3 \text{ min}$  to the set-point of the output variable using a PI controller tuned with trial and error method.

