NAME: SOSANYA IYIOLA ELIZABETH

**COLLEGE: MEDICINE AND HEALTH SCIENCES** 

**DEPARTMENT: PHARMACY** 

MATRIC NUMBER: 19/MHS11/134

**COURSE CODE: CHM 102** 

## **ASSIGNMENT**

#### 1. Classification of Alcohols

**A.** Classification based on the number of hydrogen atoms attached to the carbon atom containing the OH group

i. if 2 or 3 hydrogen atoms are attached to the carbon atom bearing the OH group, it is called a primary alcohol (1°)

ii. If 1 hydrogen atom is attached, it is called a secondary alcohol (2°)

iii. If no hydrogen atom is attached to the carbon atom, it is called a tertiary alcohol (3°)

# EXAMPLES.

Methanol CH3OH (1°)

Propan-2-ol CH3CH(OH)CH3 (2°)

**B.** Classification based on the number of hydroxyl groups they possess. Monohydric alcohol has one OH group present in the alcohol structure. Dihydric alcohols are called glycols, they have 2 hydroxyl groups present in the structures while trihydric alcohols or triols have three OH groups present in the structure of the alcohol. Polyhydric alcohols or polyols have more than 3 OH groups.

### **EXAMPLES**

Monohydric alcohol—Propanol CH3CH2CH2OH

Dihydric alcohol -Ethane1,2diol HOCH2-CH2OH

# 2.GRIGNARD SYNTHESIS OF ALKANOLS

Grignard reagents -C2H5MgBr

CH3CH2CH2CH2-C=OCH2CH2CH3 + C2H5MgBr-→C4H9C3H7C2H5-C-

OMgBr→C4H9C3H7C2H5-C-OH + Mg (OH)Br

#### 3. Industrial manufacture of ethanol

Carbohydrate such as starch are major group of natural compound that can be made to yield ethanol by the biological process of FERMENTATION. The biological catalyst enzymes found in yeast breaks down the carbohydrate molecules into ethanol to give a yield of 95%. On warming starch with malt to 60° for a specific period of time is converted in to maltose by the enzyme diastase contained in the malt.

 $2(C6H10O5) n + n H2O \rightarrow n (C12H22O11)$ 

Carbohydrate 60 °C/ diastase maltose

The maltose is broken down into glucose on addition of yeast which contains the enzymes maltase and at a temperature of  $15\,^{\circ}\text{C}$ 

C12H22O11 + H2O → 2C6H12O6

Maltose 15 °C/maltase glucose

The glucose at constant temperature of 15°C is then converted into alcohol by the enzyme zymase contained also in the yeast

C6H12O6 → 2CH3CH2OH +2CO2

Glucose 15 °C/zymase ethanol

**4.Alkanone:** Reduction of alkanone gives secondary alkanols.

CH3C2H5-C=O 
$$\rightarrow$$
 CH3C2H5CHOH (2°) alcohol LiAlH4

Alkanals: Reduction of alkanals gives primary alkanols

CH3CH2CH=O → CH3CH2CH2OH LiAlH4/H2O