NAME:ILESANMI OLUWATOBILOBA MARY MATRIC NUMBER:19/MHS11/071 CHEM 102 CLASSIFICATION OF ALKANOLS WITH EXAMPLES

- Classification based on the number of hydrogen atoms attached to the carbon atom containing the hydroxyl group: if the numbers of hydrogen atoms attached to the carbon bearing the hydroxyl group are 2 Or 3 it's called a primary alcohol. It's called a secondary alcohol when it's one hydrogen atom bearing the hydroxyl group. If no hydrogen atom is attached to the carbon bearing the hydroxyl group it's called tertiary alcohol. E.g Ch3Ch2OHethanol (1°) (2). (CH2))3C-OH-2 propane-2-ol
- 2. Classification based on the number of hydroxyl group they posses .monohydric alcohols have only one hydroxyl group present in the alcohol structure. Dihydric alcohols are also called glycols they have 2 hydroxyl group present in the alcohol structure while trihydric has 3 hydroxyl group present in the alcohol structure and polyhydric has more than 3 alcohol structure e.g. ethanol and propane 2 ol
 - 2 Grinard synthesis of alkanols

Grinard reagent-C2H5MgBr

CH3CH2CH2CH2—C=OH2CH2CH3+C2H5.MgBr

C4H9C37C2H5--C--OMgBr--→C4H9C3H7C2H5--C—OH+Mg(OH)Br

3. Industrial preparation of ethanol

Carbohydrates such as starch are major group of natural compounds that can be made to yield ethanol by the biological process of fermentation. The biological catalyst, enzymes found in yeast to break down the carbohydrates molecules into ethanol to give a yield of 95%. On warming starch with malt to 60° for a specific period of time are converted into maltose by the enzyme diastase contained in the malt.

2(C6H10O5)n + H20-----→n(C12H22011) Carbohydrate 60°diastase maltose

Maltose 15°/maltase glucose

The glucose at constant temperature of 15°c is then converted into alcohol by the enzyme contained also in yeast C6H12O6----- \rightarrow 2CH3CH2OH + 2CO2 Glucose 15°/zymase ethanol

 Alkanol reduction of Alkanone gives secondary alcohols CH3C2H5—C=O-----→CH3C2H5CHOH(2°)Alcohol LIALH4
Alkanals reduction of alkanone gives secondary alkanols CH3CH2CH=O----→CH3CH2CH2OH LIALH4/H2O