

Name:- BABATUNDE ANUOLUKAPO ORYOMI

Matric no:- 191mkhs021030.

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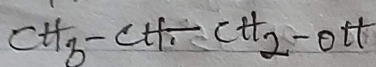
Dept:- Nursing Science

Chem 102.

100 level.

### Assignment

1. Base on the number of hydrogen atoms attached to the carbon atom containing the hydroxyl group. If the numbers of hydrogen atoms attached to the carbon atom bearing the hydroxyl group are three or two, it is called a primary alcohol (1°). If it is one hydrogen atom, it is called secondary alcohol (2°) and if no hydrogen atom is attached to the carbon atom bearing the hydroxyl group, it is called a tertiary alcohol (3°). Examples;



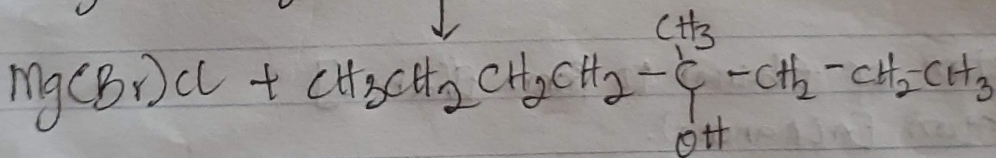
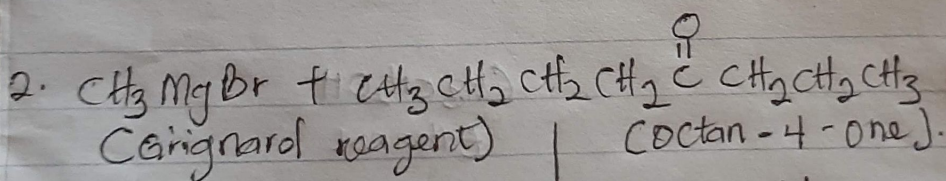
2-methylpropan-1-ol (1°)



Pentan-3-ol.

ii Based on the number of hydroxyl groups they possess. Monohydric alcohols have one hydroxyl group present in the alcohol structure. Dihydric alcohols are also called glycols have two hydroxyl groups present in the alcohol structure while trihydric alcohols or triols have three hydroxyl groups present in the structure of the alcohol. Polyhydric alcohols or polyols have more than three hydroxyl groups. Examples;  $CH_3CH_2CH_2OH$  propanol (Monohydric alcohol)

$HOCH_2CH_2OH$  ethane-1,2-diol (Dihydric alcohol).



Tertiary alcohol.

### 3. production of Ethanol.

Carbohydrates such as starch are major group of natural compounds that can be made to yield ethanol by the biological process of fermentation. The biological catalysts, enzymes found in

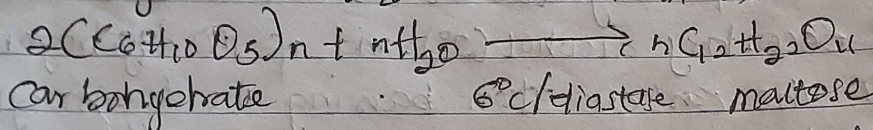
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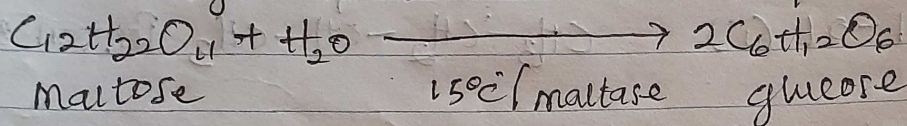
chem 102

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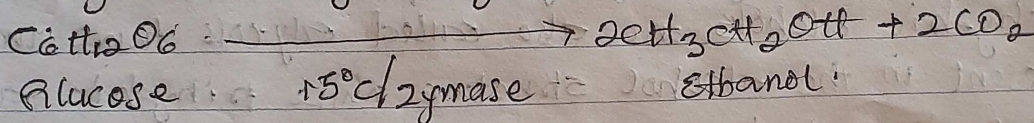
yeast break down the carbohydrate molecules into ethanol to give a yield of 95%. The starch containing materials include molasses, potatoes, cereals, rice and on warming with malt to 60°C for a specific period of time are converted into maltose by the enzyme diastase contained in the malt.



The maltose is broken down into glucose on addition of yeast which contains enzyme maltase and at a temperature of 15°C.

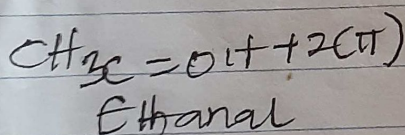


The glucose at constant temperature of 15°C is then converted into alcohol by the enzyme zymase contained also in yeast.



The product obtained from the reduction of Alkanone and Alkanals is alcohol.

Equation (Alkanals)



Reduction of ethanal to form ethanol

Equation (Alkanone)

