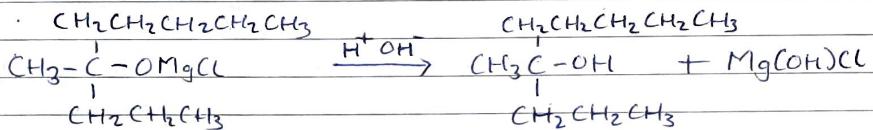
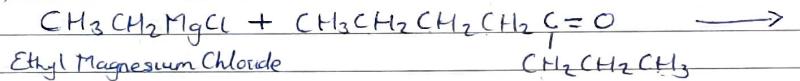


1) Alkanols can be classified based on two things;

a) They can be classified based on the number of hydroxyl groups attached to a carbon atom. In this case we have three categories; Monohydric, Dihydric and Polyhydric.

b) Secondly, they can be classified based on the number of hydrogen atoms atoms attached to the same carbon atom having a hydroxyl group. In this case there are also three categories; Primary alcohols (2 or 3 hydrogen atoms), Secondary alcohols (1 hydrogen atom) and Tertiary alcohols (No hydrogen atom).

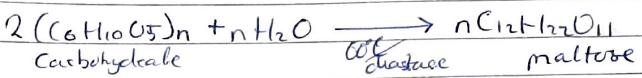
2)



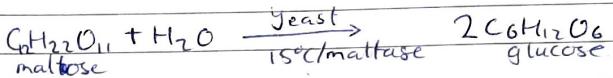
4-methyl-1-pentan-4-ol

3) Industrially, Ethanol is prepared by the fermentation of carbohydrates by the action of enzymes found in malt and yeast.

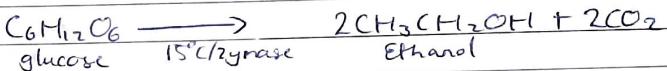
Firstly, Carbohydrates are decomposed by diastase from malt at 60°C to get maltose



Secondly, maltose is broken down by the action of maltase found in yeast which gives glucose at 15°C..

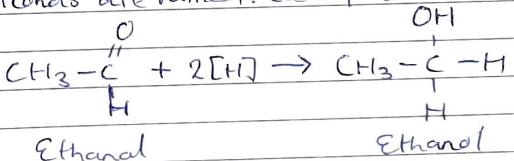


Thirdly, glucose is broken down to ethanol by the action of Zymase at 15°C.



4) For Alkanals,

When alkanals are reduced, their corresponding primary alcohols are formed. Example;



For Alkanones,

When Alkanones are reduced, their corresponding secondary alcohols are formed - Example;

