

Name: Peshwacha Phinachutai Peshwacha

MATRIC NO: 1917W01167

Assignment

① Give the IUPAC names of the following organic compounds.

① CH_3OCH_3 - Methoxyethane

② $\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_3$ - Ethoxyethane

③ $(\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2)_2\text{O}$ - Butoxybutane

④ $\text{CH}_3\text{CH}_2\text{OCH}_3$ - Methoxyethane

⑤ $\text{CH}_3\text{CH}_2\text{CH}_2\text{OCH}_2\text{CH}_3$ - Ethoxypropane

② Discuss the properties of ethers

General properties

① Physical states

At room temperature, ethers are colourless, neutral liquids with pleasant odours. The lower aliphatic ethers are highly flammable gases or volatile liquids.

② Solubility

Ethers are less soluble in water than are the corresponding alcohols. Lower molecular weight ethers such as methoxyethane and methoxypropane are fairly soluble in water since the molecules are able to form hydrogen bonds with the water molecules but as the hydrocarbon content of the molecules increases, there is rapid decline in solubility.

③ Density

Most of the simple ethers are less dense than water, although the density increases with increasing relative molecular mass and some of the aromatic ethers are in fact denser than water.

④ Boiling point

Lower molecular mass ethers have a lower boiling point than

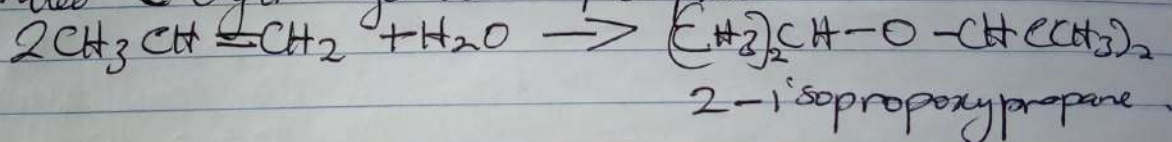
expands. The corresponding alcohols but those ethers containing alkyl radicals larger than four carbon atoms, the reverse is true. The boiling point of ethers tend to approximate those of hydrocarbons of some relative molecular mass from which it can be concluded that the molecules are not associated in the liquid phase as there are no suitably available hydrogen for association through hydrogen bonds.

⑤ Reactivity: Ethers are inert at moderate temperature. Their inertness at moderate temperature leads to their wide use as reaction media.

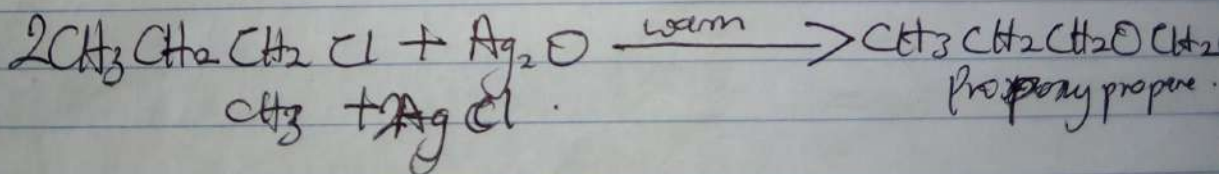
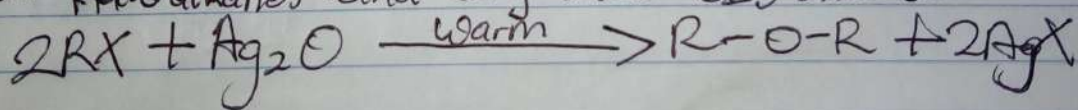
Simple ethers are not found commonly in nature but the ether linkage is present in such natural products as sugars, starches and cellulose.

③ Discuss explicitly two methods of preparing ethers and show equations of reaction.

① Controlled catalytic hydration of olefins.



② From haloalkanes and dry silver (I) oxide.



④ State three uses of ethylene oxide

① Ethylene oxide is used as an intermediate in the hydrolytic manufacture of ethylene glycol.

Ethylene oxide is used in the preparation of non-ferrous
metals, pigments, plastics, plasticizers and several textile
dyes. Ethylene oxide is used as a gaseous sterilizing agent.

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