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CHM 102

CARBOXYLIC ACID ASSIGNMENT

HUMAN ANATOMY

19/MHS03/003

- 1. Give the IUPAC names of the following compounds
 - HCOOH- Methanoic acid
 - HOOCCH₂CH₂CH₂COOH- Pentan-1,5-dioic acid
 - CH₃CH₂CHCOOH- Butanoic acid
 - HO₂C-CO₂H- Ethanedioic acid
 - CH₃(CH₂)₄COOH-Hexanoic acid
 - CH₃CH=CHCH₂CH₂COOH- Hex-4-eneoic acid
- 2. Write two industrial preparations of carboxylic acids.
 - From carbon(ii) oxide: Methanoic acid is manufactured by adding carbon(ii)oxide under pressure to hot aqueous solution of sodium hydroxide. The free carboxylic acid is liberated by careful reaction with tetraoxosulphate(Vi) acid (H₂SO₄)

NaOH
$$H_2SO_4$$
 CO------ \rightarrow HCOOH+ NaHSO $_4$

- From Ethanol: Ethanoic acid is obtained commercially by the liquid phase air-oxidation of 5% solution of ethanal to ethanoic acid using manganite(ii) ethanoate catalyst. Ethanal is obtained from ethylene.
- Dil. $H_2SO_4/HgSO_4$ $O_2/(CH_3COO)_2Mn$



- 3. With equations and brief explanations discuss the synthetic preparation of carboxylic acid
 - Oxidation of primary alcohols and aldehydes: oxidation of primary alcohols and aldehydes can be used to prepare carboxylic acids using the usual oxidizing agents (i.e K₂Cr₂O₇ or KMnO4) in acidic solution

 Carbonation of Gringard reagent: Aliphatic carboxylic acids are obtained by bubbling carbon(IV) oxide into the Gringnard reagent and then hydrolyzed with dilute acid.

Hydrolysis of nitriles(cyanides) or esters

H+

RCN+
$$2H_2O$$
------ \rightarrow RCOOH+NH₄+ H_2O /H+ reflux

4. With chemical equation only, outline the reduction, decarboxylation and esterification of carboxylic acid.

LiAIH4

Butanoic acid

butanol

Decarboxylation

fuse

Electrolysis/aq. CH3OH

• Estherification

H⁺CH3CH2CH2COOH+CH3CH2CH2OH←-→CH3CH2CH2COOCH2CH2CH3+H2