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## BETA OXIDATION

Beta oxidation is a metabolic process involving multiple steps by which fatty acid molecules are broken down to produce energy. More specifically, beta oxidation consists in breaking down long fatty acids that have been converted to acyl-CoA chains into progressively smaller fatty acyl-CoA chains. This reaction releases acetyl-CoA, FADH<sub>2</sub> and NADH, the three of which then enter another metabolic process called *citric acid cycle* or *Krebs cycle* in which ATP is produced to be used as energy. Beta oxidation goes on until two acetyl-CoA molecules are produced and the acyl-CoA chain has been completely broken down. In eukaryotic cells, beta oxidation takes place in the mitochondria, whereas in prokaryotic cells, it happens in the cytosol.

Stages of beta oxidation :

### A] DEHYDROGENATION :

In the first step, acyl-CoA is oxidized by the enzyme acyl CoA dehydrogenase. A double bond is formed between the second and third carbons (C<sub>2</sub> and C<sub>3</sub>) of the acyl-CoA chain entering the beta oxidation cycle; the end product of this reaction is trans-enoyl-CoA (trans- $\Delta^2$ -enoyl CoA). This stage uses FAD and produces FADH<sub>2</sub>, which will enter the citric acid cycle and form ATP to be used as energy.

### B] HYDRATION :

In the second step, the double bond between C<sub>2</sub> and C<sub>3</sub> of trans- $\Delta^2$ -enoyl-CoA is hydrated, forming the end product L- $\beta$ -hydroxyacyl CoA, which has a [hydroxyl group](#) (OH) in C<sub>2</sub>, in place of the double bond. This reaction is catalyzed by

another enzyme. The trans alkene is then hydrated with the help of Enoyl-CoA hydratase

**C] OXIDATION :** In the third step, the hydroxyl group in C2 of L- $\beta$ -hydroxyacyl CoA is oxidized by  $\text{NAD}^+$  in a reaction that is catalyzed by 3-hydroxyacyl-CoA dehydrogenase. The end products are  $\beta$ -ketoacyl CoA and  $\text{NADH} + \text{H}^+$ .  $\text{NADH}$  will enter the citric acid cycle and produce ATP that will be used as energy. The alcohol of the hydroxyacyl-CoA is then oxidized by  $\text{NAD}^+$  to a carbonyl with the help of Hydroxyacyl-CoA dehydrogenase.  $\text{NAD}^+$  is used to oxidize the alcohol rather than [FAD] because  $\text{NAD}^+$  is capable of the alcohol while [FAD] is not.