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CHM102

MECHATRONICS ENGINEERING

Assignment on Stereochemistry and
functional Group

19/ENG05/017

Name the functional groups present in each of the following molecules

$\text{CH}_2 = \text{C}(\text{OH})\text{HCHO}$: Alkene ($=$), Alcohol (OH), Alkanal (CHO)

" $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3$: Amines (RNH_2), Alkanones (RCOR)

" $\text{CH}_3\text{C} = \text{CHCH}(\text{OH})\text{CHO}$: Alkene ($=$), Alcohol (OH), Alkanal (CHO)

2. A 0.856g sample of pure (2R, 3R)-tartaric acid was diluted to 10cm^3 with water and placed in a 1.0dm polarimeter tube. The observed rotation at 20°C was $+1.0^\circ$. Calculate the specific rotation of (2R, 3R)-tartaric acid.

Solution

$$\text{mass} = 0.856\text{g}$$

$$\text{Temperature} = 20^\circ\text{C}$$

$$\text{Volume} = 10\text{cm}^3$$

$$l = 1\text{dm}$$

$$\text{Observed rotation } \alpha = +1.0 \text{ (a Dextrorotary)}$$

$$\text{Specific rotation} = ?$$

But,

$$\text{Specific rotation} = \frac{\text{Observed rotation} (^\circ)}{\text{Concentration (g/cm}^3) \times \text{Path length (dm)}}$$

$$\alpha_{\text{sp}}^T = \frac{\alpha}{c \cdot l}$$

$$\text{Concentration} = \frac{\text{mass}}{\text{Volume}}$$

$$= \frac{0.856}{10}$$

$$\text{Concentration} = 0.0856\text{g/cm}^3$$

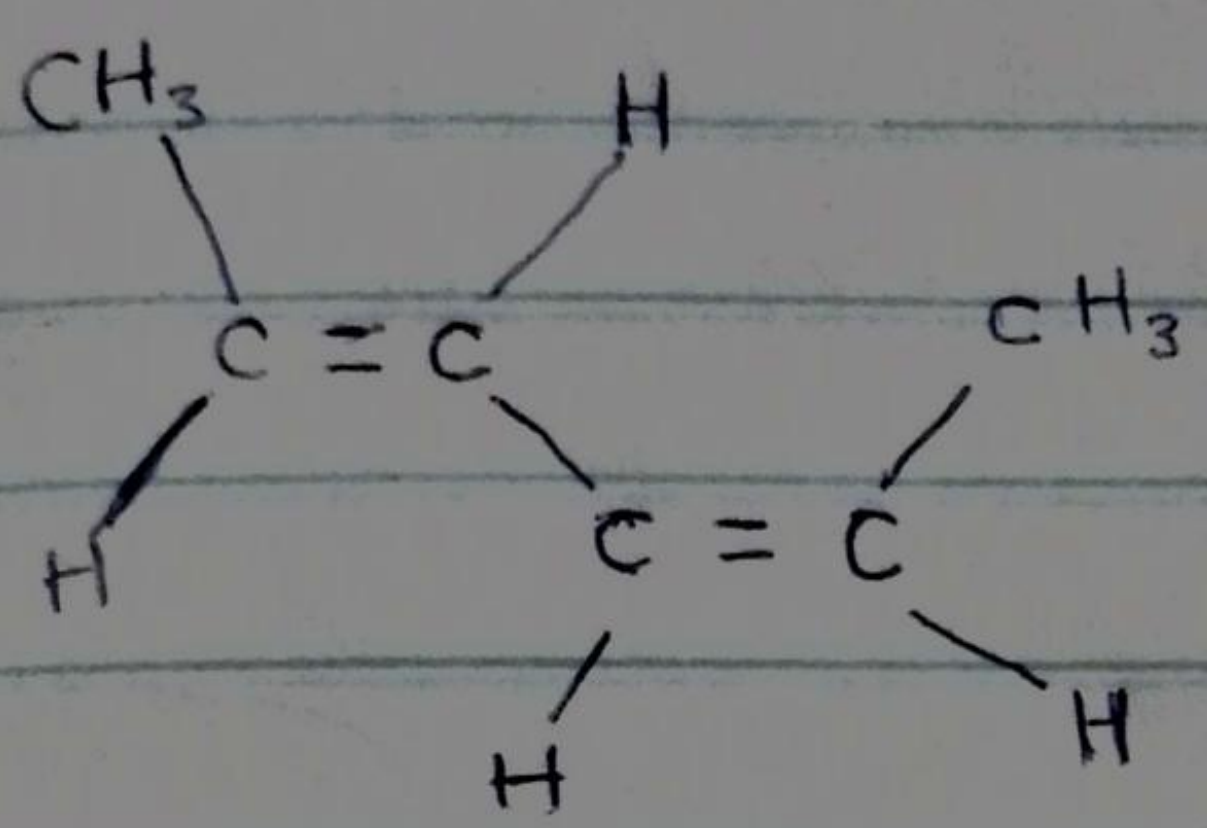
$$\alpha_{\lambda}^{20^{\circ}\text{C}} = \frac{+1.0^{\circ}}{0.0856 \cdot l}$$

$$\alpha_{\lambda}^{20^{\circ}\text{C}} = 11.68^{\circ} \text{g}^{-1} \text{cm}^3 \text{dm}^{-1}$$

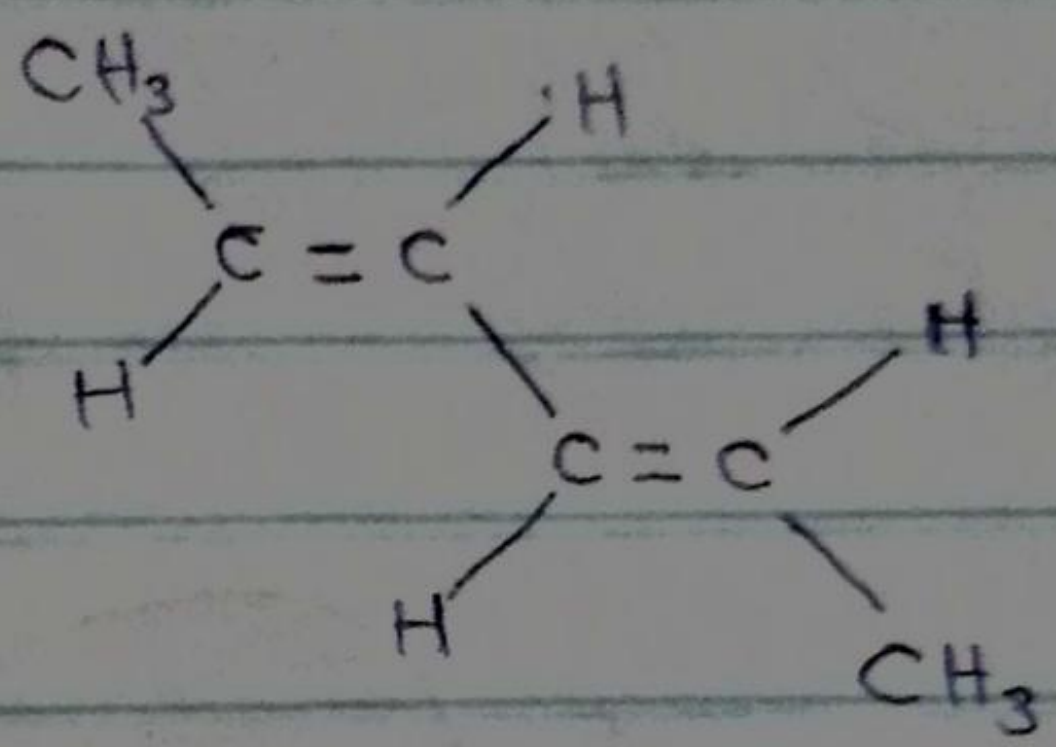
3. Draw the possible geometric isomers [where possible] for each of the following compounds

1. Hexa-2,4-diene

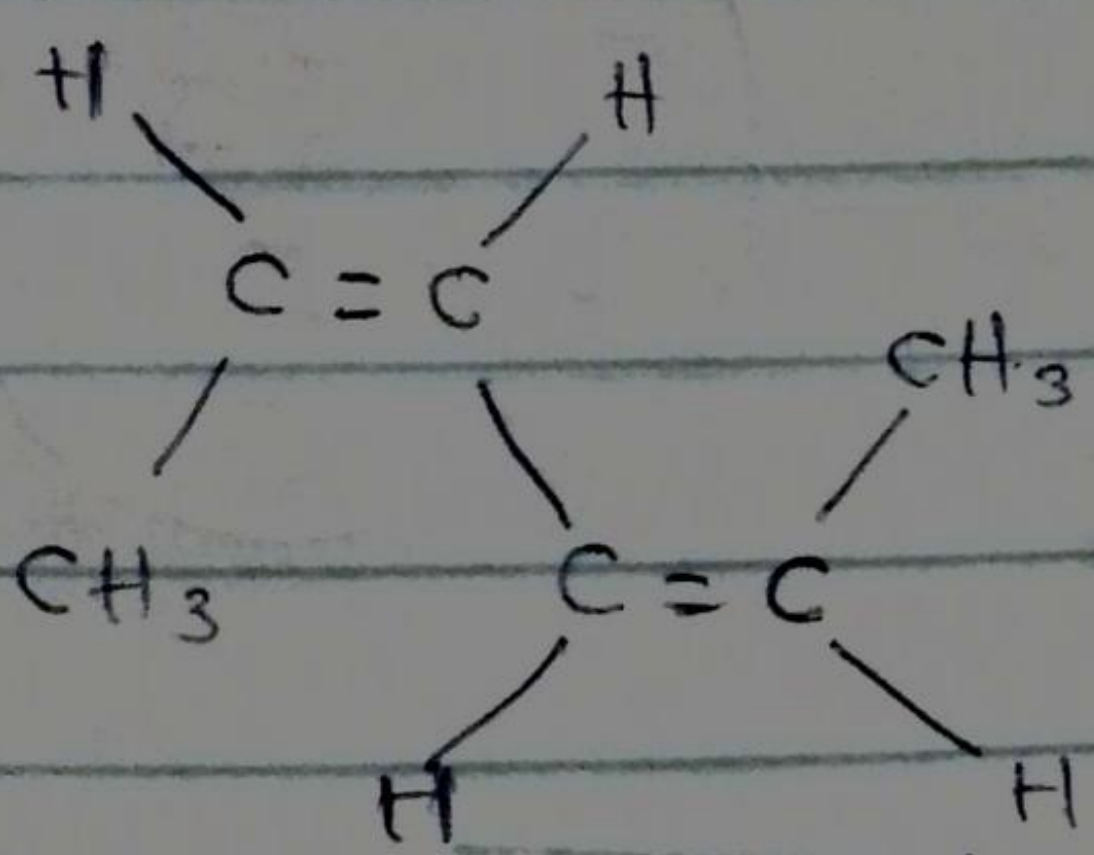
The possible geometric isomers are Three (3) isomers



~~trans, cis~~ - Hexa-2,4-diene
 $(2E, 4Z)$ - Hexa-2,4-diene



trans, trans - Hexa-2,4-diene
 $(2E, 4E)$ Hexa-2,4-diene



~~cis, cis~~ Hexa-2,4-diene
 $(2Z, 4Z)$ Hexa-2,4-diene

2,3-Dimethylbut-2-ene

