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COURSE CHEMISTRY 102

MATRIC NO 19/MHSO/194

DEPT MEDICINE AND SURGERY (MBS)

THIRD ASSIGNMENT

① Give the IUPAC names of the following compounds

Answer

a) HCOOH \longrightarrow Methanoic acid

b) $\text{HOOCCH}_2\text{CH}_2\text{CH}_2\text{COOH}$ \longrightarrow Pentan-1, 5-dioic acid

c) $\text{CH}_3\text{CH}_2\text{CH}_2\text{COOH}$ \longrightarrow Butanoic acid

d) $\text{HO}_2\text{C}-\text{CO}_2\text{H}$ \longrightarrow Ethanedioic acid

e) $\text{CH}_3[\text{CH}_2]_4\text{COOH}$ \longrightarrow Hexanoic acid

f) $\text{CH}_3\text{CH}=\text{CHCH}_2\text{CH}_2\text{COOH}$ \longrightarrow Hex-4-enoic acid

② Discuss briefly the physical properties of carboxylic acids under the following headings

i) PHYSICAL APPEARANCE

All simple aliphatic carboxylic acids up to C_{10} are liquids at room temperature. Most other carboxylic acids are solids at room temperature although anhydrous carboxylic acid (acetic acid) also known as glacial ethanoic acid freezes to an ice-like solid below the room temperature.

ii) Boiling Points

Boiling points increases with increasing relative molecular mass. Aromatic carboxylic acids are crystalline solids and have higher melting points than their aliphatic counterparts of comparable relative molecular mass.

iii) Solubility

Lower molecular mass carboxylic acid with up to four carbon atoms in their molecules are soluble in water.

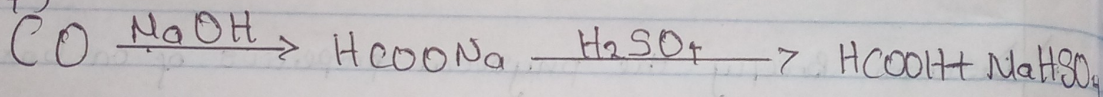
thos largely due to their ability to form hydrogen bonds with water molecules. The water solubility of the acids decreases as the relative molecular mass increases because the structure becomes relatively more hydrocarbon in nature and hence covalent. All carboxylic acids are soluble in organic solvents.

3. Write two industrial preparations of carboxylic acids.

Answer

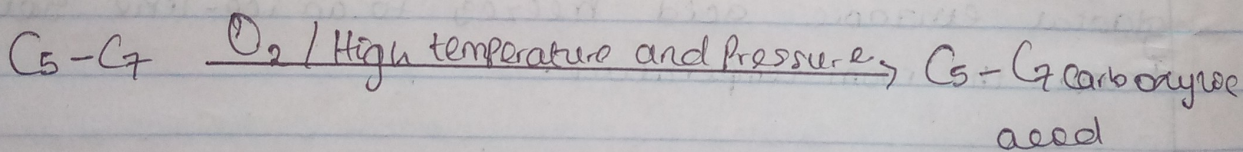
i) From Carbon (II) oxide

Methanoic acid (formic acid) is manufactured by adding carbon (II) oxide under pressure to hot aqueous solution of sodium hydroxide. The free carboxylic acid is liberated by careful reaction with tetraoxosulphate (VI) acid (H_2SO_4)



ii) From Petroleum

liquid phase air oxidation of $C_5 - C_7$ alkanes, obtainable from petroleum at high temperature and pressure will give $C_5 - C_7$ carboxylic acids with methanoic, propanoic and butanedioic acid as by-products.

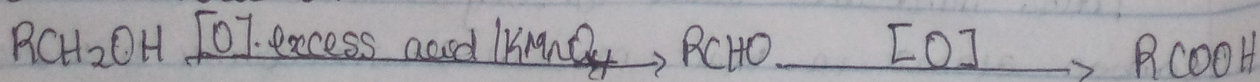


(4) With equations and brief explanation discuss the synthetic preparation of carboxylic acid.

Answer

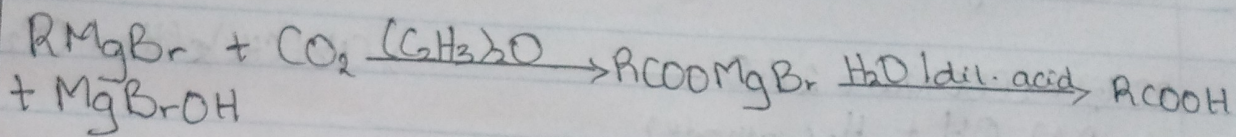
i) Oxidation of primary alcohols and aldehydes

Oxidation of primary alcohols and aldehydes can be used to prepare carboxylic acids using the usual oxidizing agents (i.e. $K_2Cr_2O_7$ or $KMnO_4$) in acidic solution



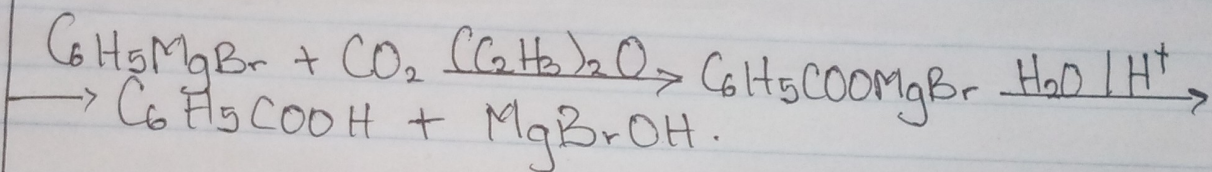
(ii) Carbonation of Grignard reagent

Aliphatic carboxylic acids are obtained by bubbling carbon (iv) oxide into the Grignard reagent and then hydrolyzed with dilute acid.

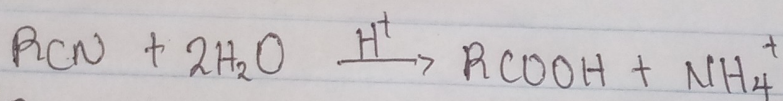


R may be 1°, 2°, 3° aliphatic alkyl or aryl radical

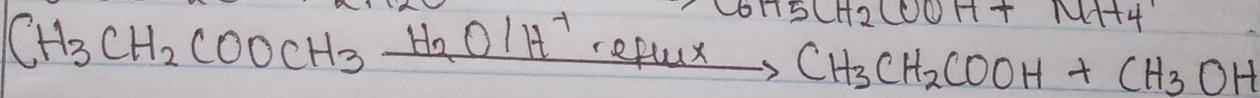
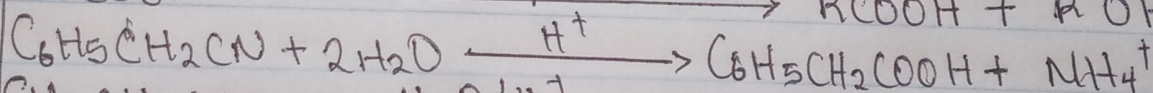
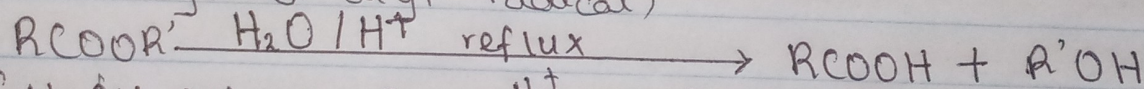
In the preparation of benzoic acid, the reagent is added to solid carbon (iv) oxide (dry ice) which also serves as coolant to the reaction mixture.



(iii) Hydrolysis of nitriles (cyanides) or esters



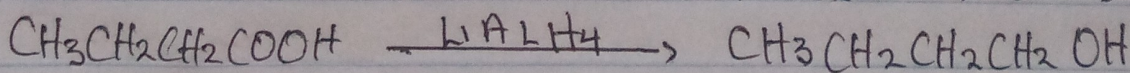
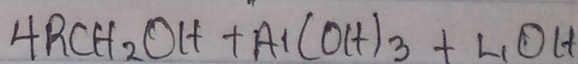
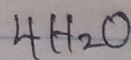
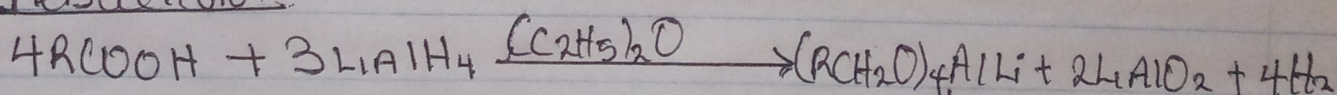
(R = alkyl or aryl radical)



(5) With chemical equation only, outline the reduction, decarboxylation and esterification of carboxylic acid.

Answer

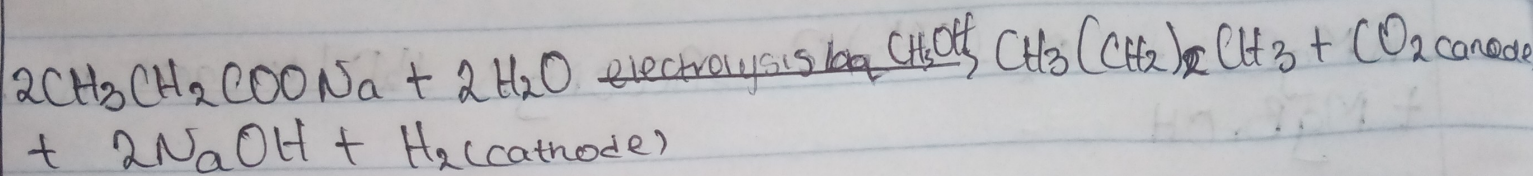
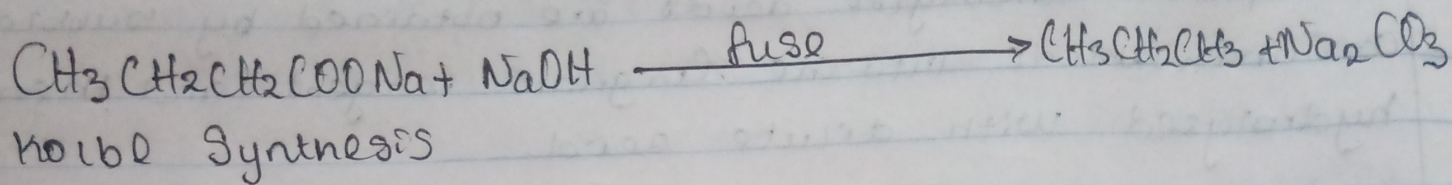
Reduction



Butanoic acid

Butanol

(vi) Decarboxylation



(vii) Esterification

