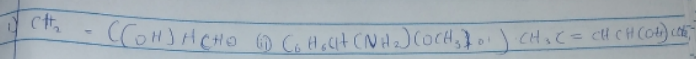


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Stereochemistry and functional group

1) Name the functional group present in the following molecule



| Molecules | Functional groups |
|--|--|
| i) $\text{CH}_2 = (\text{OH})\text{HCHO}$ | Hydroxyl group (-OH) Aldehyde (-CHO) |
| ii) $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)(\text{OCH}_3)$ | Amine (-NH ₂) Carbonyl Group (CO) |
| iii) $\text{CH}_3\text{C} = \text{CHCH}(\text{OH})\text{CHO}$ | Aldehyde (-CHO) Hydroxyl group (-OH) |

2) A 0.856 g sample of pure (2R, 3R) tartaric acid and was diluted to 10 cm³ with water and placed in a 1.0 dm Polarimeter tube; the observed rotation at 20°C was +10°. Calculate the specific rotation of (2R, 3R) - tartaric acid.

$$[\alpha] = \frac{\alpha}{cL}$$

where $[\alpha]$ = specific optical rotation
 α = observed rotation = +10°
 c = concentration in g/ml = ?
 L = Path length (in dm) = 1.0 dm

The concentration is always measured in g/ml. From the question the mass is in g. So therefore, the volume of water will be converted from 10 cm³ to ml, where;

$$1 \text{ lme} = 1000 \text{ cm}^3 \quad 1 \text{ cm}^3 = 1 \text{ ml} = \frac{1000^3 \times 1 \text{ ml}}{1000^3}$$

$$1 \text{ m} = 1 \text{ cm}^3 \quad 10 \text{ cm}^3 = ? \quad 1 \text{ cm}^3$$

∴ Conc in g/ml = 10/g ml

$$\therefore [\alpha] = \frac{\alpha}{d \cdot l} = \frac{1.0}{10 \times 10} = \frac{1}{10}$$

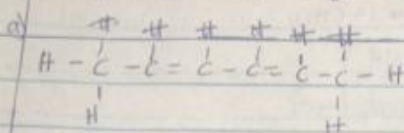
$$[\alpha] = \frac{1}{10} = 0.1^\circ$$

\therefore Specific rotation of (2R,3R)-tartaric acid is 0.1°

3 Draw the possible geometric isomers (where possible) for each of the following compounds

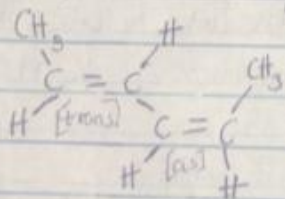
i) Hexa-2,4-diene ii) 2,3-Dimethyl but-2-ene

i) Hexa-2,4-diene $(CH_2=CH=CH=CH=CH_2)$

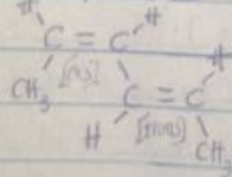


Hexa-2,4-diene

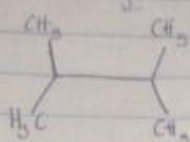
b) Trans-1, Cis-4-dimethyl but-2-ene



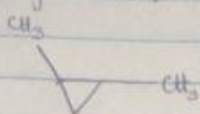
Cis-1, trans-4-dimethyl but-2-ene



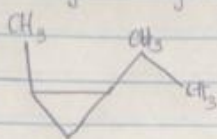
i. 2,3-Dimethyl but-2-ene



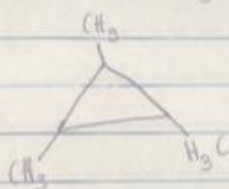
ii. 1-methyl-trans-2-ethyl cyclopropane



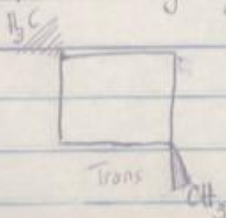
b) Cis-1-methyl-2-ethyl cyclopropane



c) 1-(is-2-trans-3-trimethyl) cyclopropane



d) Trans-1,3-dimethyl cyclobutane

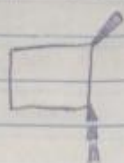


e Cis-1,2,3-trimethyl cyclopropane



f Trans-1,2-dimethyl cyclobutane

3



g Cis-1,3-dimethyl cyclobutane

