

UKWELLA JOANNA IBEJO

MBBS

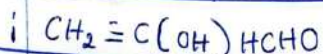
100 Level

19/MHSC1/418

CHM 102

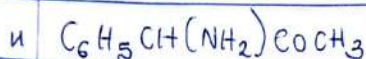
Assignment: Stereochemistry and Functional group.

1 Name the functional groups present in each of the following



$\Rightarrow$  Groups present

- Aldehyde group (formaldehyde) ( $\text{CH}_2\text{O}$ )
- Alkanol group ( $-\text{OH}$ ) / Hydroxyl group
- Alkene group ( $\text{C}=\text{C}$ )



$\Rightarrow$  Groups Present

- Amine group ( $-\text{NH}_2$ )
- Ketone group ( $\text{C} \begin{smallmatrix} \text{O} \\ \parallel \\ \text{CH}_3 \end{smallmatrix}$ )
- Phenyl group ( $\text{C}_6\text{H}_5$ )



- Alkene group ( $\text{C}=\text{C}$ )
- Alkanal group ( $-\text{C} \begin{smallmatrix} \text{O} \\ \parallel \\ \text{H} \end{smallmatrix}$ ) Aldehyde group

2 A 0.856g sample of pure (2R, 3R)-tartaric acid was dissolved in 10 cm<sup>3</sup> with water placed in a 1.0 dm polarimeter tube. The rotation at 20°C was +1°. Calculate the specific rotation of (2R, 3R) tartaric acid.

Soln

Specific rotation =  $\frac{\text{Observed rotation}}{\text{con. in g/cm}^3 \times (\text{path length in cell})}$

$$[\alpha] = \frac{\alpha}{c \times l}$$

Observed rotation =  $+1.0^\circ$

$$C = \frac{0.856 \text{ g}}{10 \text{ cm}^3}$$

$$= 0.0856 \text{ g/cm}^3$$

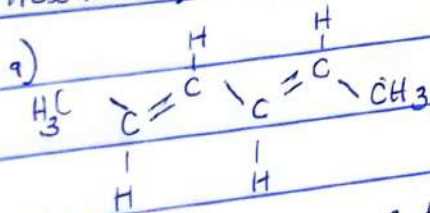
$$l = 1 \text{ dm}$$

$$\text{Specific rotation} = \frac{+1.0^\circ}{0.0856 \text{ g/cm}^3 \times 1 \text{ dm}}$$

$$\text{Specific rotation} = 11.68 \text{ g}^{-1} \text{ cm}^3 \text{ dm}^{-1}$$

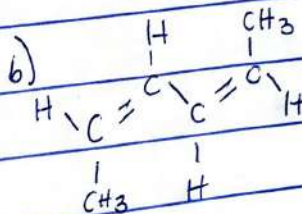
3 Draw the possible geometric isomers for the following compound

1) Hexa-2,4-diene

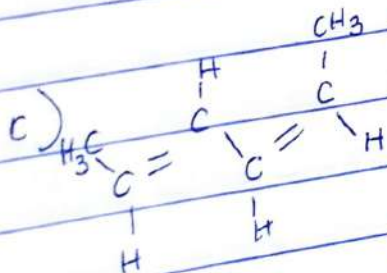


(2E, 4E)-hexa-2,4-diene

(cis, cis)-hexa-2,4-diene

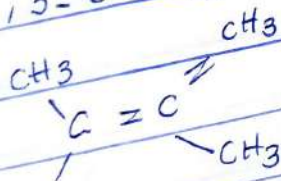


(2Z, 4Z)-hexa-2,4-diene



(2Z, 4E)-hexa-2,4-diene

2) 2,3-Dimethylbut-2-ene



no geometric isomers.