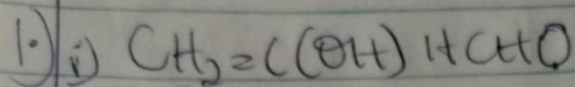
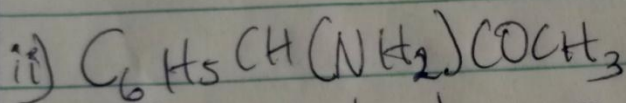


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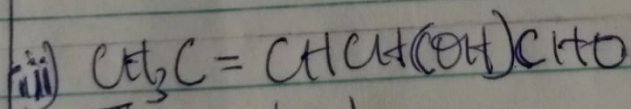
The functional groups:

- Alkene functional group $-\overset{|}{\text{C}}=\overset{|}{\text{C}}-$
- Hydroxyl group $-\text{OH}$
- Aldehyde functional group $-\text{CHO}$



The functional groups:

- Amine functional group $-\text{NH}_2$
- Ketone functional group $-\text{COR}$



The functional groups:

- Alkene functional group $-\overset{|}{\text{C}}=\overset{|}{\text{C}}-$
- Hydroxyl functional group $-\text{OH}$
- Aldehyde functional group $-\text{CHO}$

2) Recall;

$$[\alpha]_D^T = \frac{\alpha}{l \times c}$$

where

l = length of sample

c = mass (g/dm) or (g/mol)
volume

α = observed rotation

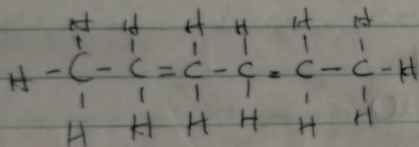
$$S_D = \frac{1.0}{1.0 \times \left(\frac{0.856}{10}\right)}$$

$$S_D = 11.68$$

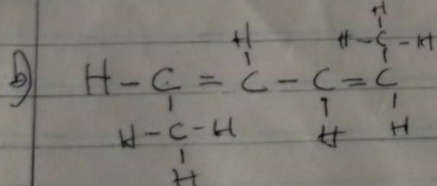
$$S_D = 11.68$$

(I)

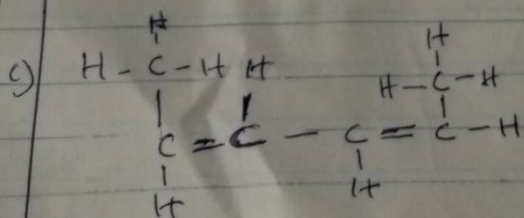
3) Hexa-2,4-diene



trans-trans-Hexa-2,4-diene



(i)-(i) Hexa-2,4-diene



trans(i) or (i)-trans-Hexa-2,4-diene

3) ii) 2,3-
It d
groups C=C
making d
isomers.

3) ii) 2,3-dimethyl but-2-ene

It does not have geometric isomers because there are two identical groups (CH_3) attached to the same carbon on the double bond making it impossible to separate it into (c) and trans geometric isomers.