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1.Alcohols are very important organic compounds. Discuss briefly their classification and give one example each.

ANSWER

a. CLASSIFICATION BASED ON NUMBER OF HYDROGEN ATOMS ATTACHED TO CARBON ATOMS: If the numbers of hydrogen attached to the carbon atom bearing the hydroxyl group are two or three, it is called a primary alchohol(1°), . If it is one hydrogen atom, it is called secondary alchohol(2°). and if no hydrogen atom is attached to the carbon atom bearing the hydroxyl group, it is called a tertiary alchohol(3°).

Examples:CH₃CH₂CH₂CH₂OH(Butanol(1°)

b. CLASSIFICATION BASED ON THE NUMBER OF HYDROXYL GROUPS THEY POSSESS: Monohydric alchohols have one hydroxyl group present in the alchohol structure. Dihydric alchohols have two hydroxyl groups present in the alchohol structure while trihydric alchohols have three hydroxyl groups present in the structure of the alchohol. Polyhydric alchohols have more than three hydroxyl groups. Examples: CH₃CH₂CH₂OH Propanol(Monohydric alchohol)

2. Discuss the solubility of alcohols in water, organic solvents

ANSWER

Lower alchohols with up to three carbon atoms in their molecules are soluble in water because these lower alchohols can form hydrogen bond with water molecules. The solubility of water decreases with increasing molecular mass. All monohydric alchohols are soluble in organic solvents. The solubility of simple and polyhydric alchohols is largely due to their ability to form hydrogen bonds with water molecules.

3. Show the three steps in the industrial manufacture of ethanol. Equations of reaction are mandatory

ANSWER

I. Conversion of starch: Starch containing material including rice, Yam etc on warming with malt to 60° C for a specific period of time are converted into maltose by the enzyme diastase contained in malt.

 $2(C_6H_{12}O_5)_n + nH_2O_{---->} nC_{12}H_{22}O_{11}$

Carbohydrate 60°C/ diastase maltose

II. Breakdown of maltose: The maltose is broken down into glucose on addition of yeast which contains the enzyme maltase and at a temperature of 15° C.

 $C_{12}H_{22}O_{11} + H_2O_{-----> 2C_6H_{12}O_6}$

Maltose 15°c/ maltase glucose

III.Conversion of glucose: The glucose at constant temperature of 15°C is then

converted into alchohol by the enzyme zymase contained in yeast.

 $C_6H_{12}O_6 - - - - - > 2CH_3CH_2OH + 2CO_2$

Gluose 15°c/ zymase Ethanol

4. Show the reaction between butylmagnesium chloride and 2-methyl propanal

Reaching between Britzl magnesium chloride & 2-methyl popu-4. CH3 CH2 CH2 CH2 mgCl + CH3 CH3 CH-C=0 ________ CH3 CH3 CH2 CH2 -C-Omgcl _______ CH3 CH3 CH -C-Omgcl _______ CH3 CH3 CH -C-OH

7. Show the reduction reaction of 2-methylpropanal

H2 (CH3)2CHCHO ----->(CH3)2CHCH2OH LiAlH4 8. Propose a scheme for the conversion of propan-1-ol to propan-2-ol

ANSWER

 $-H_{2}O$ $CH_{3}CH_{2}CH_{2}OH + H_{2}SO_{4} -----> CH_{3}CH_{2}CH_{2}OSO_{3}H -----> Propanol -H_{2}SO_{4}$ $CH_{3}CH_{2}CH_{2}OSO_{3}H -----> CH_{2}=CHCH_{3} + H_{2}O -----> CH_{2}CHOHCH_{3}$ Prop-2-ene Propan-2-ol