

Saved Photos

Specific rotation of the Sample = $\frac{1}{0.856} \times 1$
 $= \frac{1}{0.856} \times 1$
 $= 11.68^\circ \text{g}^{-1} \text{cm}^3 \text{dm}^{-1}$

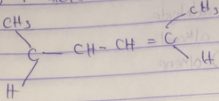
3) Draw the possible geometric isomers (where possible) for each of the following compounds

(a) Hexa-2,4-diene

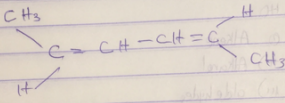
ii) 2,3-Dimethyl but-2-ene

Answers

(a) Hexa-2,4-diene

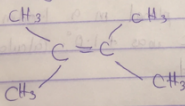


cis-Hexa-2,4-diene



trans-Hexa-2,4-diene

ii) 2,3-Dimethyl but-2-ene



Geometric isomers is not possible for 2,3-dimethyl but-2-ene

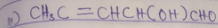
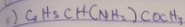
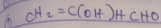
Name: Isha Favour Fernandes

Matric no: 1911M15081065

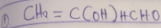
Department: Nursing

Course Code: Chem 102

Q) Name the functional groups present in each of the following molecules



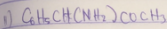
Answers



Functional group: i) aldehyde

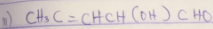
ii) alcohol

iii) alkene



Functional group: i) amides

ii) ketones



Functional Group: i) Alkene

ii) Alcohol

iii) aldehydes

Q) A 0.856g sample of pure (2R,3R)-tartaric acid was diluted to 100ml with water and placed in a 1.0dm polarimeter tube. The observed rotation at 20°C was +1.0°. Calculate the specific rotation of (2R,3R)-tartaric acid.

Answers

Observed rotation = +1.0°

$$\text{Conc.} = \frac{0.856}{1.0} = 0.856 \text{ g cm}^{-3}$$

length of sample cell (polarimeter) = 1.0 dm

$$\therefore \text{Specific rotation} = \frac{\text{observed rotation (degrees)}}{\text{Concentration (g cm}^{-3}) \times \text{path length of sample cell in dm}}$$