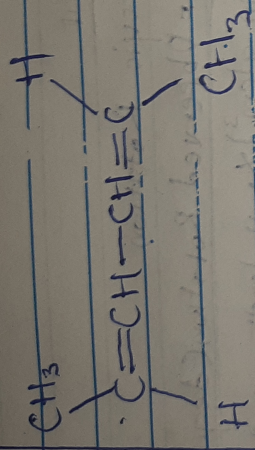


Electrostatics is the branch of Physics which deals with the study of electric charges at rest.

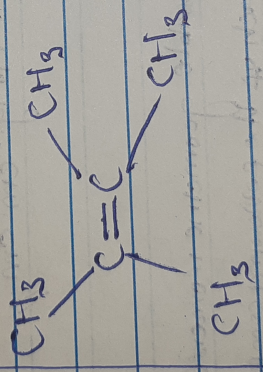
| | |
|---|--|
| <p>Oybonna Justice Nnaoma 1911MH8011286</p> | <p>2. folw</p> |
| <p>MIBB Stereochemistry</p> | <p>Specific Rotation</p> |
| <p>I Name the Functional group in each of the following molecules</p> | <p>= Observed Rotation (degree) $[\alpha]_{\text{conc. in g/cm}^3} \times \text{path length of sample in dm}$</p> |
| <p>1 $\text{CH}_2 = \text{C}(\text{OH})\text{HCHO}$</p> | <p>Specific Rotation = $\frac{0.886 \text{ g} / 10 \text{ cm}^2 \times 1}{1}$</p> |
| <p>Functional Group: Alkene, Alcohol, Aldehyde</p> | <p>$= 11.0 \text{ g}^{-1} \text{ cm}^3 \text{ dm}^{-1}$</p> |
| <p>11. $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3$</p> | |
| <p>Functional Group: Amide, Ketone</p> | |
| <p>111. $\text{CH}_3\text{C}=\text{CHCH}(\text{OH})\text{CHO}$</p> | <p>3) Draw the possible geometric isomers (where possible) for each of the following compound. 2-Hexa-2,4-diene ($\text{CH}_3\text{CH}=\text{CH}-\text{CH}=\text{CH}-\text{CH}_3$)</p> |
| <p>Functional group: Alkene, Alcohol, Aldehyde</p> | <p>$\begin{array}{c} \text{H} \\ \\ \text{C}=\text{CH}-\text{CH}=\text{C} \\ \quad \quad \quad \\ \text{CH}_3 \quad \quad \quad \text{CH}_3 \end{array}$</p> |
| | <p>cis-Hexa-2,4-diene</p> |

electric charge is a physical property of when near



Trans Hexa-2,4-diene

ii) 2,3-Dimethylbut-2-ene
 $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)_2$



There are no isomers for this compound.