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Course: General Chemistry II (CHM 102)

1. Name the functional groups present in each of the following molecules



Functional groups present

- a. $\text{C}=\text{C}$ (Alkene)
- b. $-\text{OH}$ (Hydroxyl group)
- c. $-\text{CHO}$ (Alkanal)



Functional groups present

- a. Phenyl group
- b. $-\text{NH}_2$ (Amine)
- c. $-\text{C}=\text{O}$ (Alkanone)



Functional groups present

- a. $\text{C}=\text{C}$ (Alkenes)
- b. $-\text{OH}$ (Hydroxyl group)
- c. $-\text{CHO}$ (Alkanal)

2. A 0.856g sample of pure (2R, 3R) Tartaric acid was diluted to 10cm^3 with water and placed in a 1.0 polarimeter tube. The observed rotation at 20°C was $+1.0^\circ$. Calculate the specific rotation of (2R, 3R) Tartaric acid.

Mass of pure (2R,3R)-Tartaric acid = 0.856g

Volume = 10cm³

Observed rotation = +1.0°

Path length = 1.0dm

Specific rotation = ??

Concentration in g/cm³ = $\frac{0.856}{10} = 0.0856 \text{ g/cm}^3$

Specific Rotation = $\frac{\text{Observed rotation}}{(\text{Concentration}) \times (\text{path length})}$

Specific rotation = $\frac{+1.0^\circ}{(0.0856) \times (1.0 \text{ dm})}$

= +11.68 g⁻¹cm³dm⁻¹

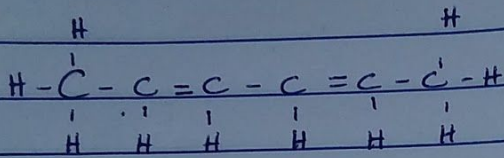
∴ ⇒ Specific Rotation

= +11.68 g⁻¹cm³dm⁻¹

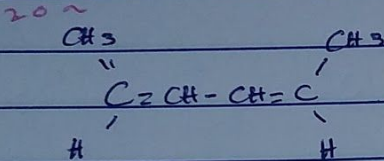
3. Draw the geometric isomers (where possible) for each of the following compounds

- Hexan-2,4-diene
- 2,3-dimethylbut-2-ene

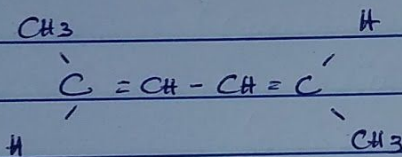
1. Hexan-2,4-diene



Isomers

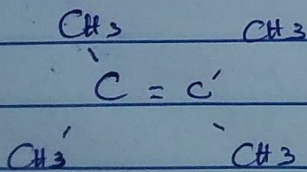
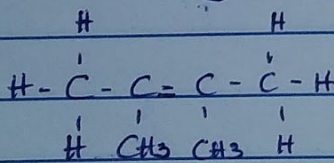


Cis-2,4-hexadiene



Trans-2,4-hexadiene

ii) 2,3-dimethylbut-2-ene



It has no cis-trans isomerism as all substituents are identical