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MATRIC NUMBER: 19/ENG04/002
ELECTRICAL/ELECTRONICS DEPARTMENT
CHM 102

1. $\text{CH}_2=\text{C}(\text{OH})\text{HCHO}$ ---- ALKANOL AND ALDEHYDE
 $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3$ --- AMINE AND KETONE
 $\text{CH}_3\text{C}=\text{CHCH}(\text{OH})\text{CHO}$ --- ALKANOL AND ALDEHYDE

$$2 \cdot [\alpha]_D^{25} \rightarrow \frac{\alpha}{Lc}$$

$$0.856\text{g in } 10\text{cm}^3 \rightarrow 10\text{ml of solution}$$

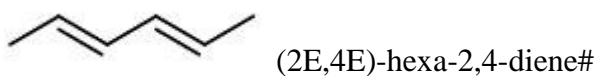
$$[\alpha]^{20^\circ} \rightarrow \frac{+1.0}{(1.00\text{dm})(0.0856)} \rightarrow +11.68^\circ$$

3.(i) Hexa-2,4-diene

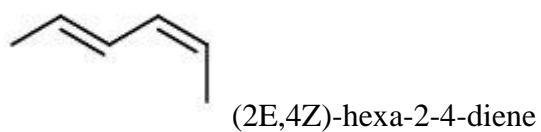
The four possibilities are (E, E), (E, Z), (Z, E), and (Z, Z).

However, (E, Z) and (Z, E) are identical.

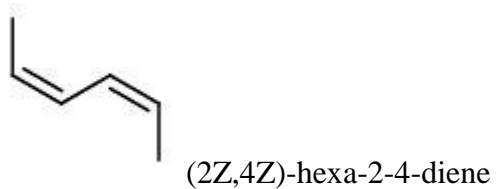
(2E,4E)-hexa-2,4-diene



(2E,4Z)-hexa-2,4-diene



(2Z,4Z)-hexa-2,4-diene



(ii). 2,3-Dimethyl but-2-ene

