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MATRIC NO: 19/MHS01/226

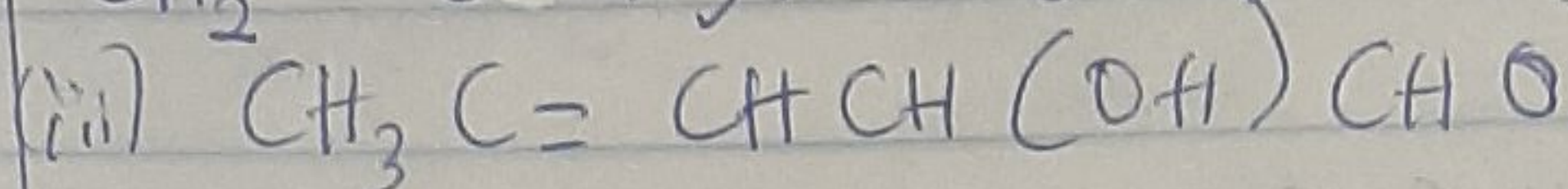
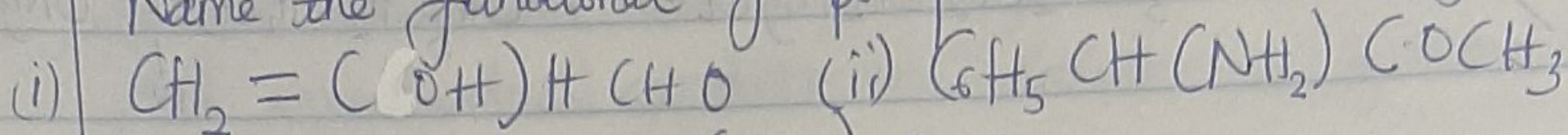
COLLEGE: MEDICINE AND HEALTH SCIENCES

DEPARTMENT: MEDICINE AND SURGERY

CHM 102 ASSIGNMENT ON STEREOCHEMISTRY AND FUNCTIONAL GROUP

Question 1:

Name the functional groups present in each of the following molecules



ANSWER:

- i $\text{CH}_2 = \text{C}(\text{OH})\text{HCHO} \longrightarrow -\text{COH}$ aldehyde
 ii $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3 \longrightarrow -\text{NH}_2$ amine
 iii $\text{CH}_3\text{C} = \text{CHCH}(\text{OH})\text{CHO} \longrightarrow -\text{COH}$ aldehyde

2. A 0.856 g sample of pure (2R, 3R)-tartaric acid was diluted to 10 cm^3 with water and placed in a 1.0 dm polarimeter tube. The observed rotation at 20°C was $+1.0^\circ$. Calculate the specific rotation of (2R, 3R)-tartaric acid

Solution

Specific rotation, $\alpha_\lambda^T = \frac{\alpha}{C \cdot L}$

$\alpha = +1.0^\circ$, $C = \frac{0.856 \text{ g}}{10 \text{ cm}^3} = 0.0856 \text{ g cm}^{-3}$, $L = 1.0 \text{ dm}$

$\alpha_\lambda^T = \frac{+1.0}{0.0856 \text{ g cm}^{-3} \times 1.0 \text{ dm}} = 11.68^\circ \text{ g}^{-1} \text{ cm}^3 \text{ dm}^{-1}$

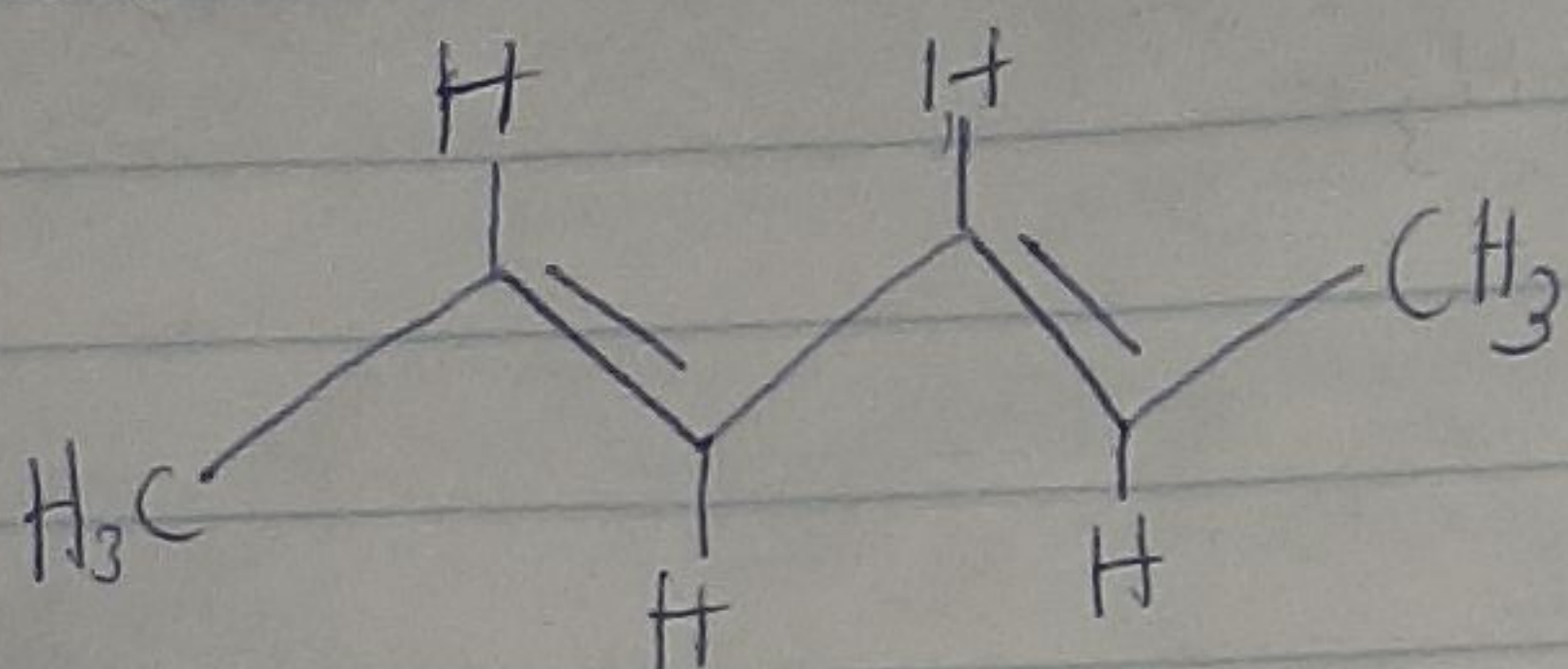
\therefore The Specific rotation of (2R, 3R)-tartaric acid is $11.68^\circ \text{ g}^{-1} \text{ cm}^3 \text{ dm}^{-1}$

3. Draw the possible geometric isomers (where possible) for each of the following compounds. (i) Hexa-2,4-diene

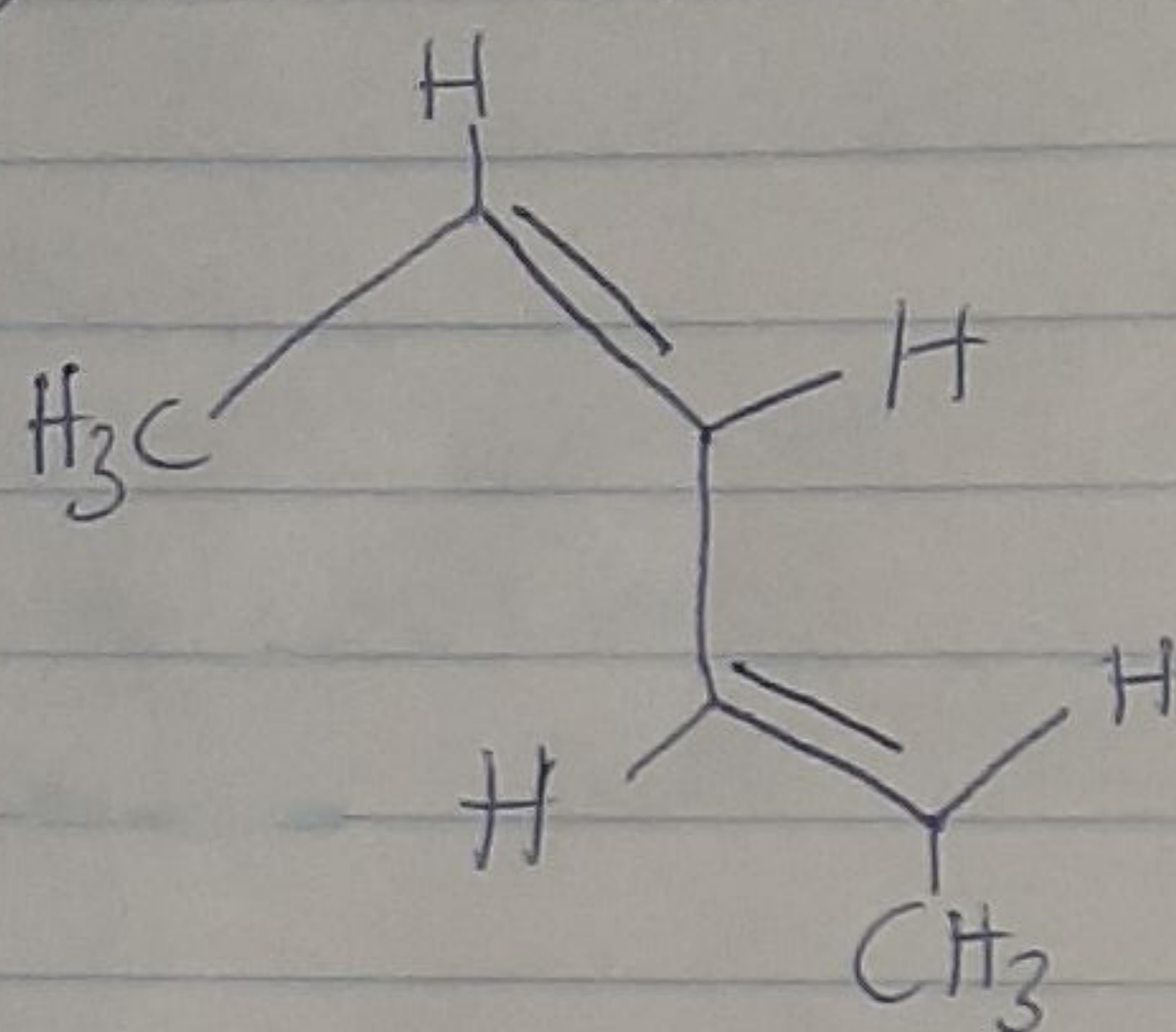
(ii) 2,3-Dimethylbut-2-ene

i. Hex-2,4-diene

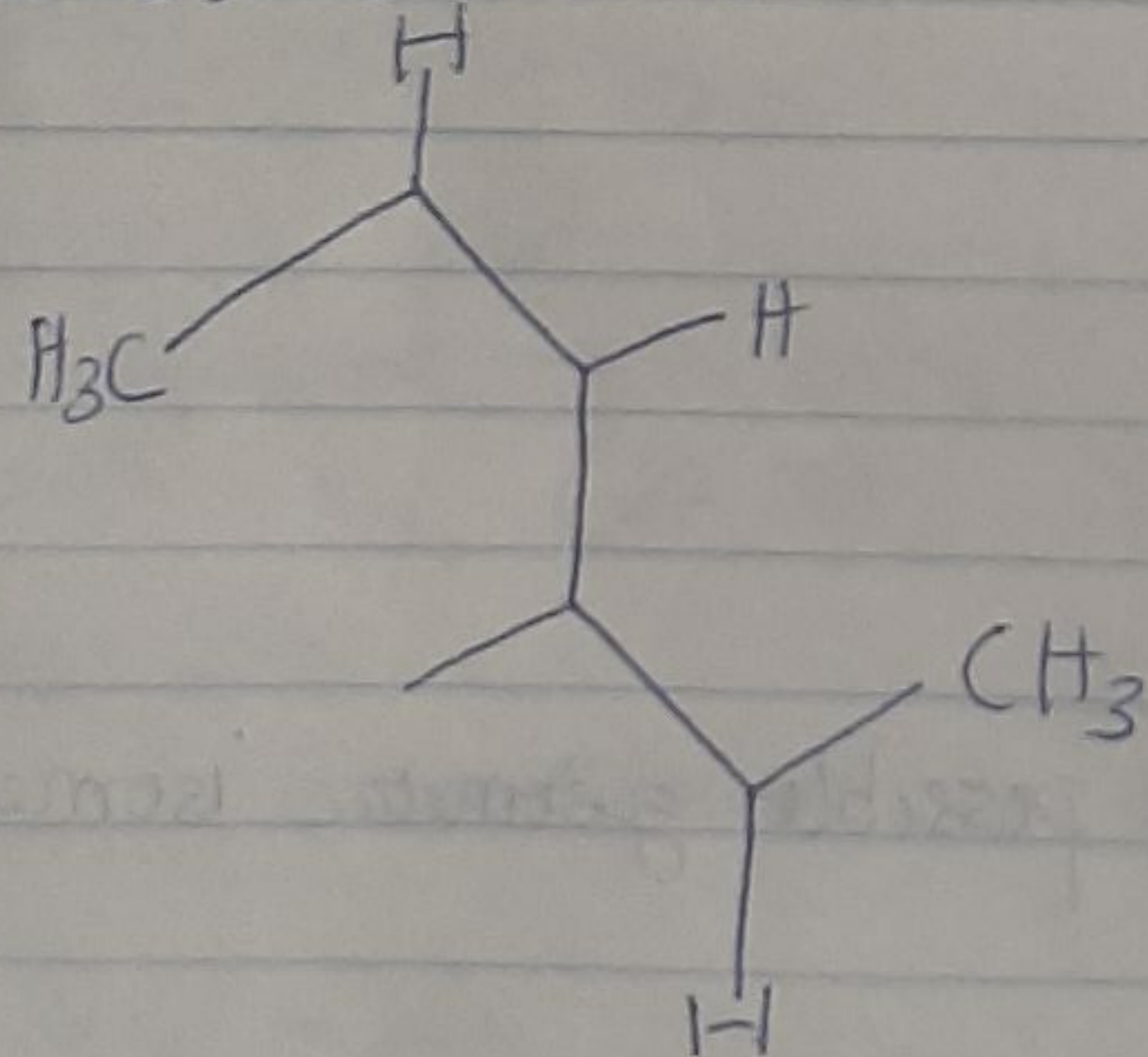
a) Trans-Trans Isomer



b) Cis-trans Isomer



c) Cis-cis Isomer



ii. 2,3-Dimethylbut-2-ene has no possible geometric isomer