

ADULTANO-050 PATRICKIA OLAMIDE

191MH601023

MEDICINE AND SURGERY

CHEM 102 ASSIGNMENT

1.) Name the functional groups present in each of the following molecules.

- a) $\text{CH}_2 = \text{C}(\text{OH})\text{CHO}$: Alkene group, Aldehyde group, hydroxyl group.
- b) $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3$: Amine group, Acetyl group.
- c) $\text{CH}_3\text{C} = \text{CHCH}(\text{OH})\text{CHO}$: Alkene group, Hydroxyl group, Aldehyde group.

2. A 0.856g sample of pure (2R,3R) tartaric acid was dissolved to 10cm^3 with water and placed in a 1.0dm polarimeter tube. The observed rotation at 20°C was $+1.0^\circ$. Calculate the specific rotation of (2R,3R)-tartaric acid.

Solution:

0.856g in 10cm^3 of solution

$$1\text{cm}^3 = 1\text{ml}$$

$$10\text{cm}^3 = 10\text{ml}$$

$$0.856 \times 10 = 0.0856\text{gmL}^{-1}$$

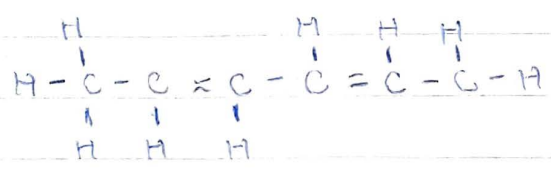
$$[\alpha]_D^{20} \approx \alpha / d, \text{ so } [\alpha] = 1.0 / (1.00\text{dm})(0.0856) = 1.0 / 0.0856 = +11.68$$

Specific rotation of (2R,3R) tartaric acid

3. Draw the possible geometric isomers (where possible) for each of the following compounds: a) Hexa-2,4-diene b) 2,3-Dimethylbut-2-ene

a.

a. Hexa-2,4-diene



B.) 2,3-Dimethylbut-2-ene

