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19/MHS01/088

CHM 102

D Name the functional groups present.

i) $\text{CH}_2 = \text{C}(\text{OH})\text{HCHO}$ — Aldehyde ($\text{C}-\text{CHO}$)
— Hydroxyl group ($\text{C}-\text{OH}$)

ii) $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3$ — Carbonyl group ($\text{C}-\text{CO}$)
— Amine ($\text{C}-\text{NH}_2$)

iii) $\text{CH}_3\text{C} = \text{CHCH}(\text{OH})\text{CHO}$ — Aldehyde ($\text{C}-\text{CHO}$)
— Hydroxyl group ($\text{C}-\text{OH}$)

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$$\text{Specific rotation} = \frac{\text{observed rotation (in degree)}}{\text{conc in g cm}^{-3} \times \text{path length (in dm)}}$$

$$\begin{aligned} \text{conc. g/cm}^3 &= \frac{0.856 \text{ g}}{10 \text{ cm}^3} \\ &= 0.0856 \text{ g/cm}^3 \end{aligned}$$

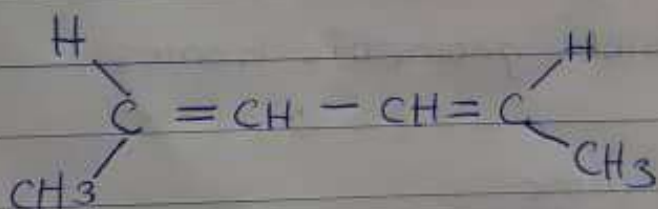
$$\text{Observed rotation} = +1.0^\circ$$

$$\text{path length of sample} = 1 \text{ dm}$$

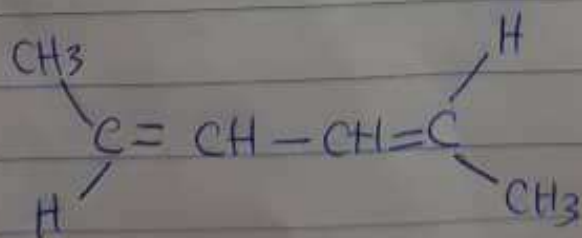
$$\text{specific rotation} = \frac{+1}{[0.0856] \times 1}$$

$$= 11.68 \text{ g}^{-1} \text{ cm}^3 \text{ dm}^{-1}$$

3) i Hexa-2,4-diene

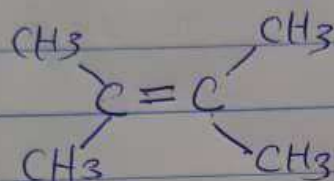


Cis Hexa-2,4-diene



Trans Hexa-2,4-diene.

ii) 2,3-dimethylbut-2-ene



Geometric isomerism is not possible 2,3-dimethylbut-2-ene.