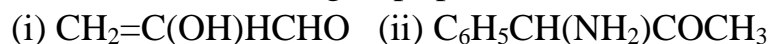


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Course: CHM102

ASSIGNMENT

1. Name the functional groups present in each of the following molecules.



Answer



Functional group present are;

- Double bond chain (Alkene)
- OH (Hydroxyl group)
- $\begin{array}{c} \text{C} = \text{O} \\ | \\ \text{H} \end{array}$ (Alkanol)



Functional group present are;

- Phenol group with double bonds
- Amine (NH_2)
- Alkanone / Ketone ($\begin{array}{c} \text{C} = \text{O} \\ \diagdown \\ \text{R} \end{array}$)



Functional group present are;

- Double bond ($=$)
- $\begin{array}{c} \text{C} = \text{O} \\ | \\ \text{H} \end{array}$ (Alkanol)
- OH (Hydroxyl group)

2. A 0.856g sample of pure (2R, 3R) – tartaric acid was diluted to 10cm³ with water and placed in a 1.0dm polarimeter tube. The observed rotation at 20⁰C was +1.0⁰. Calculate the specific rotation of (2R, 3R)-tartaric acid.

Answer

Mass of tartaric acid = 0.856g

Vol. of water diluted in = 10cm³

Recall;

$$\text{Mass Conc. (C)} = \frac{\text{mass}}{\text{vol.}} = \frac{0.856\text{g}}{10\text{cm}^3}$$

Vol. of Polarimeter tube = 1dm³

Therefore; Length of tube = 1dm

Observed rotation (α) = ?

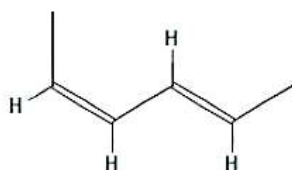
$$\begin{aligned}(\alpha) &= \frac{\alpha}{C \times L} \\&= \frac{+1.0^0}{0.0856\text{gcm}^{-3} \times 1\text{dm}}\end{aligned}$$

$$(\alpha) = \frac{+1.0^0}{0.0856} = +11.6822^0$$

3. Draw the possible geometric isomers (where possible) for each of the following compounds;

(i) Hexa-2,4-diene (ii) 2, 3-Dimethylbut-2-ene

Answer



(i) Hexa-2,4-diene – has only 3 isomers

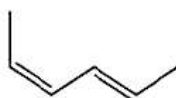
Isomers

trans-trans isomer



(2E,4E)-hexa-2,4-diene

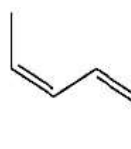
cis-trans isomer



(2E,4Z)-hexa-2,4-diene

(2Z,4E)-hexa-2,4-diene

cis-cis isomer



(2Z,4Z)-hexa-2,4-diene

- (ii) 2,3 dimethylbut-2-ene – does not have geometric isomers because there are two identical group attached to the same carbon of the double bond

