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17/ENGO1/030

Chemical Engineering
ENG382 Assignment

$$u_t - c u_{xx} = 0 \quad \text{for } 0 < x < 1$$

$$\frac{du}{dt} - \frac{cd^2u}{dx^2} = 0$$

$$\frac{du}{dt} = cd^2u/dx^2$$

$$\frac{u_{ij+1} - u_{ij}}{\Delta t} = c \cdot u_0 + c_{ij} - 2u_0 c_{ij} + u_{i-1,j}$$

$$u_{ij+1} - u_{ij} = \frac{c \Delta t}{\Delta x^2} [u_0 + c_{ij} - 2u_0 c_{ij} + u_{i-1,j}]$$

$$u(x, 0) = x^4$$

$$\Delta x = 0.2m, \Delta t = 0.002 \text{ days}$$

for initial conditions:

$$\text{At } x = 0 \rightarrow x^4 = 0$$

$$\text{At } x = 0.2 \rightarrow 0.2^4 = 1.6 \times 10^{-3}$$

$$\text{At } x = 0.4 \rightarrow 0.4^4 = 0.0256$$

$$\text{At } x = 0.6 \rightarrow 0.6^4 = 0.1296$$

$$\text{At } x = 0.8 \rightarrow 0.8^4 = 0.4096$$

$$\text{At } x = 1 \rightarrow 1^4 = 1$$

$$\therefore u_{ij+1} = u_{ij} + r [u_0 + c_{ij} - 2u_0 c_{ij} + u_{i-1,j}]$$

$$u_{ij+1} = r u_{i-1,j} + (1 + 2r) u_{ij} + r u_{i+1,j} \quad \text{--- (1)}$$

For $i=1$ when $j=1$

$$u_{1,1} = 0.5 u_{0,0} + 0.5 u_{2,0}$$

$$= 0.5(0) + 0.5(0.0256)$$

$$u_{1,1} = 0.0128$$

when $i = 2, j = 0$

$$\begin{aligned} u_{2,11} &= 0.5(u_{2,10}) + 0.5(u_{3,10}) \\ &= 0.5(1.6 \times 10^{-3}) + 0.5(0.1296) \\ u_{2,11} &= 0.0656 \end{aligned}$$

when $i = 3, j = 0$

$$\begin{aligned} u_{3,11} &= 0.5(u_{2,10}) + 0.5(u_{4,10}) \\ &= 0.5(0.0256) + 0.5(0.4096) \\ u_{3,11} &= 0.2176 \end{aligned}$$

when $i = 4, j = 0$

$$\begin{aligned} u_{4,11} &= 0.5(u_{2,10}) + 0.5(u_{6,10}) \\ &= 0.5(0.0256) + 0.5(0.4096) \\ u_{4,11} &= 0.5648 \end{aligned}$$

For $j = 2$

when $j = 1, i = 1$

$$\begin{aligned} u_{1,12} &= 0.5(0_{11}) + 0.5(u_{2,1}) \\ &= 0.5u_{11} + 0.5u_{2,1} = 0 + 0.5(0.656) \\ u_{1,12} &= 0.328 \end{aligned}$$

when $j = 1, i = 2$

$$\begin{aligned} u_{2,12} &= 0.5(u_{11}) + 0.5(u_{3,11}) \\ &= 0.5(0.0128) + 0.5(0.2176) \\ u_{2,12} &= 0.1152 \end{aligned}$$

when $j = 1, i = 3$

$$\begin{aligned} u_{3,12} &= 0.5(u_{0,11}) + 0.5(u_{4,11}) \\ &= 0.5(0.0656) + 0.5(0.5648) \\ u_{3,12} &= 0.3152 \end{aligned}$$

when $j = 1, i = 4$

$$\begin{aligned} u_{4,12} &= 0.5(u_{2,11}) + 0.5(u_{5,11}) \\ &= 0.5(0.2176) + 0.5(1) \\ u_{4,12} &= 0.6088 \end{aligned}$$

for $j=2$

$$\text{when } i=1 \rightarrow U_{113} = 0.5(U_{011}) + 0.5(U_{212}) \\ = 0 + 0.5(0.1152) = 0.576 \\ = 0.0576.$$

$$\text{when } i=2 \rightarrow U_{213} = 0.5(U_{112}) + 0.5(U_{312})$$

$$= 0.5(0.0320) + 0.5(0.3152) \\ = 0.174$$

$$\text{when } i=3 \rightarrow U_{313} = 0.5(U_{113}) + 0.5(U_{412})$$

$$= 0.5(0.1152) + 0.5(0.6088) \\ = 0.362$$

$$\text{when } i=4 \rightarrow U_{413} = 0.5(U_{114}) + 0.5(U_{312})$$

$$= 0.5(0.8152) + 0.5(r)$$

$$U_{413} = 0.6576$$

for $J=3$

$$\text{when } i=1 \rightarrow U_{114} = 0.5(U_{013}) + 0.5(U_{213}) \\ = 0.5(0) + 0.5(0.174) \\ = 0.084$$

$$\text{when } i=2 \rightarrow U_{214} = 0.5(U_{113}) + 0.5(U_{313})$$

$$= 0.5(0.0576) + 0.5(0.362) \\ = 0.2098$$

$$\text{when } i=3 \rightarrow U_{314} = 0.5(U_{213}) + 0.5(U_{413})$$

$$= 0.5(0.174) + 0.5(0.6576) \\ = 0.4158$$

$$\text{when } i=4 \rightarrow U_{414} = 0.5(U_{313}) + 0.5(U_{513})$$

$$= 0.5(0.362) + 0.5(1) \\ = 0.681$$

for $J = 4$

$$\text{when } i=1 \quad u_{1,5} = 0.5(u_{1,4}) + 0.5(u_{2,4}) \\ = 0.5(0.2098) + 0 \\ = 0.1049$$

$$\text{when } i=2 \quad u_{2,5} = 0.5(u_{1,4}) + 0.5(u_{3,4}) \\ = 0.5(0.687) + 0.5(0.4158) \\ = 0.2514$$

$$\text{when } i=3 \quad u_{3,5} = 0.5(u_{2,4}) + 0.5(u_{4,4}) \\ = 0.5(0.4454) + 0.5(0.687) \\ = 0.5679$$

$$\text{when } i=4 \quad u_{4,5} = 0.5(u_{3,4}) + 0.5(u_{5,4}) \\ = 0.5(0.4158) + 0.5(1) \\ = 0.7079$$

TABLE

Δt	j	Temp(K)					
0.1	5	0	0.1049	0.2514	0.4454	0.7079	1
0.05	4	0	0.087	0.2098	0.4158	0.681	1
0.06	3	0	0.0576	0.174	0.362	0.6576	1
0.04	2	0	0.0328	0.1152	0.3152	0.6008	1
0.02	1	0	0.0128	0.0656	0.2176	0.5648	1
0	0	0	0.0016	0.028	0.1296	0.4096	1
Δx		0	0.2	0.4	0.6	0.8	1
	i	0	1	2	3	4	5

x/t	0	0.2	0.4	0.6	0.8	1
0	0	0.0016	0.0256	0.1296	0.4096	1
0.02	0	0.0128	0.0656	0.2176	0.5648	1
0.04	0	0.0328	0.1152	0.3152	0.6088	1
0.06	0	0.0576	0.174	0.362	0.6576	1
0.08	0	0.087	0.2098	0.4158	0.681	1
0.1	0	0.1049	0.2514	0.4454	0.7079	1

Plot of Temperature Profile of Rod

c	1
r=c*(Δt/Δx2)	0.5
Δt	0.02
Δx2	0.04

