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MATRIC NO: 19/mhs04/002

ASSIGNMENT TITLE: Stereochemistry and Functional Group

COURSE TITLE: General Chemistry II

COURSE CODE: CHM 102

1. Name the functional groups present in each of the following molecules

(i) $\text{CH}_2=\text{C}(\text{OH})\text{HCHO}$ (ii) $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3$ (iii) $\text{CH}_3\text{C}=\text{CHCH}(\text{OH})\text{CHO}$

Answer

(i) $\text{CH}_2=\text{C}(\text{OH})\text{HCHO}$

– COH Aldehydes

- OH Alkanols

(ii) $\text{C}_6\text{H}_5\text{CH}(\text{NH}_2)\text{COCH}_3$

– NH_2 Amines

– RCOR' Alkanones

(iii) $\text{CH}_3\text{C}=\text{CHCH}(\text{OH})\text{CHO}$

– OH Alkanols

- COH Aldehydes

2. A 0.856 g sample of pure (2R, 3R)-tartaric acid was diluted to 10cm^3 with water and placed in a 1.0 dm polarimeter tube. The observed rotation at 20°C was +1.00. Calculate the specific rotation of (2R, 3R)-tartaric acid.

$$\text{Specific rotation} = \frac{\text{Observed rotation (degrees)}}{\text{Concentration (g/cm}^3\text{) x path length of sample cell in (dm)}}$$

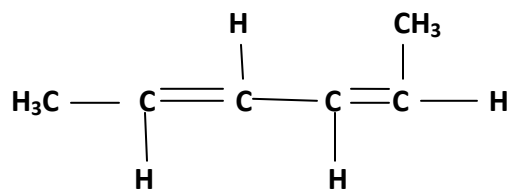
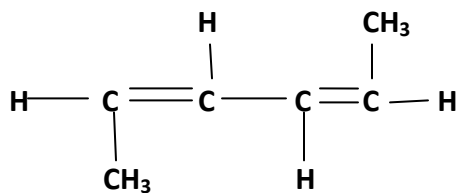
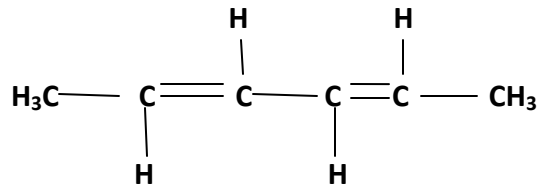
$$\begin{aligned}\text{Specific rotation of (2R, 3R)-tartaric acid} &= \frac{20^\circ\text{C}}{(0.856\text{g}/10\text{cm}^3) \times 1.0\text{dm}} \\ &= 20 / 0.0856 \\ &= 233.64^0\text{g}^{-1}\text{cm}^3\text{dm}^{-1}\end{aligned}$$

3. Draw the possible geometric isomers (where possible) for each of the following compounds:

(i) Hexa-2,4-diene (ii) 2,3-Dimethylbut-2-ene

Answer

(i) Hexa-2,4-diene



(ii) 2,3-Dimethylbut-2-ene

It cannot show Geometric Isomerism.