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18/mhs06/019

Medical Lab Science.

1. Fat soluble vitamins and Water soluble vitamins. The biochemical importance of vitamins is that it acts as a regulator of cell and tissue growth and differentiation. It provides a hormone like function, regulating mineral metabolism for bones and other organs. Some vitamins function as enzyme cofactors or precursors.

2. Thiamin helps in breaking down glucose for energy and acts as a coenzyme in the metabolism of a maintenance agent (it helps in maintaining the cognitive ability of the branch-chain amino acids). Riboflavin is a part of two co-enzymes (flavin mononucleotide and adenosine dinucleotide). It helps in the oxidation-reduction during energy production within the metabolic pathways. It helps in extracting energy needed for metabolism, it helps in the counter free-radical damage.

3. Nucleotide Nomenclature! It contains three characteristic components: a nitrogenous base, a pentose and one or more phosphate groups.

Nucleoside Nomenclature! Nucleosides are glycosylamines that can be thought of as nucleotides without a phosphate group. It consists of a nucleobase (nitrogenous base) and five carbon sugar ribose.

Nucleic acids Nomenclature! They are composed of nucleotides, which are the monomers made of three components: a 5 carbon sugar, a phosphate group and a nitrogenous base. If the sugar is a compound ribose the polymer is RNA; if the sugar is derived from ribose as deoxyribose, the polymer is DNA.

4. The role of vitamin A in the visual cycle is specifically related to the retinal form. Within the eye, 11-cis-retinal is bound to the protein "opsin" to form rhodopsin (in rods and iodopsin (cones) at condensed lysine residues. As light enters the eye, the 11-cis-retinal is isomerized to the all-trans form. The all-trans retinal dissociates from the opsin in a series of steps called photobleaching. This isomerization includes a nervous signal along the optic nerve to the visual center of the brain. After separating from opsin, the all-trans-retinal is recycled and converted back to the 11-cis-retinal form by a series of enzymatic reactions. In addition, some of the all-trans-retinal may be

may be converted to all-trans-retinal form and then transported with an interphotoreceptor retinal-binding protein (IRBP) to the pigment epithelial cells. Further esterification into all-trans-retinol within the pigment epithelial cells to be reused when needed. The final stage is conversion of 11-cis-retinal re-binding to opsin to reform rhodopsin (visual purple) in the retina. Rhodopsin is needed to see in low light contrast as well as for night vision. Kuhse showed that rhodopsin in the retina is only regenerated when the retina is attached to retinal pigmented epithelium, which provides retinal. It is for this reason that a deficiency in vitamin A will inhibit the reformation of rhodopsin and will lead to one of the first symptoms, night blindness.

6. The named vitamin is vitamin D which is a hormone. The skin is responsible for producing vitamin D. During exposure to sunlight, ultraviolet radiation penetrates into the epidermis and photolyzes provitamin D3 to previtamin D3. Provitamin D3 can either isomerize to vitamin D3 or be photolyzed to lyxsterol and tachysterol. Vitamin D is also sensitive to sunlight and is photolyzed to 5,6-trans vitamin D3. The epidermis possesses receptors for $1,25-(OH)_2-D_3$. $1,25-(OH)_2-D_3$ inhibits the proliferation of cultured keratinocytes and induces their death. The skin is the site for the synthesis of vitamin D from the sun and a target tissue for active metabolites. When the skin is exposed to sunlight it makes vitamin D from cholesterol. The sun's ultraviolet B rays hit cholesterol in the skin cells, providing the energy for vitamin D synthesis to occur.

7. If nucleic acids are exposed to moderate concentration of alkali it causes ^{deprotonation} ~~deprotonation~~ causing the DNA to be denatured. If they are exposed to high concentration of alkali it will induce a hydrolysis of phosphodiester bonds which would be cleaved to smaller fragments. If nucleic acids are exposed to ^{high acids} ~~acids~~ the DNA is deprotonated at purines which causes DNA melting and the DNA sequence is lost and when they are exposed to extremely high acids the phosphodiester bonding of the DNA is disrupted which cleaves the DNA into nucleosides and nucleotides.

8. According to James Watson (1928): - The DNA is a double-stranded helical molecule. It consists of two sugar phosphate backbones on the outside, held together by hydrogen bonds between pairs of nitrogenous base on the inside. The bases are of four types (A, C, G and T).

: Pairing always occurs
- Francis Crick (1919)
either strand contains
the entire molecule
provide a genetic code

8 RNA

1. RNA contains ribose
2. RNA is a single molecule
3. RNA is ^{not} stable under conditions
4. RNA directly codes for amino acids and acts as a template between DNA and protein
5. RNA uses adenine, uracil, cytosine, and guanine for its bases
6. RNA is related to UV damage

10 a) Nucleotide

- a) is the main energy source
- b) Nucleotides also form the backbone of DNA
- c) They are used in many biological processes
- d) It is found in many processes including transporting

Pairing always occurs between A & T, and C and G -
 - Francis Crick (1916-2004) realized that these pairing rules meant that either strand contained information necessary to make a new copy of the entire molecule and that the aperiodic order order of bases might provide a "genetic code"

RNA	DNA
1. RNA contains sugar deoxy-ribose	1. DNA contains sugar ribose
2. RNA is a single-stranded molecule	DNA is a double-stranded molecule
3. RNA is ^{not} stable under alkaline conditions	DNA is stable under alkaline conditions
4. RNA directly codes for amino acids and acts as a messenger between DNA and ribosomes to make proteins	DNA is responsible for storing and transferring genetic information.
5. RNA uses thymine adenine, uracil, cytosine and guanine for its base pairing	DNA uses thymine, cytosine and guanine for its base pairing.
6. RNA is relatively resistant to UV damage	DNA is susceptible to UV damage.

10. A Nucleotide can be a base in another molecule such as (ATP) which is the main energy molecule of the cell

5) Nucleotides not only make up the building blocks of life but also form different molecules that function to make life possible

a) They are found in co-enzymes like ~~and~~ NAD and NADP which are used in many chemical reactions that play different roles in metabolism

d) It is found in cyclic AMP as messenger molecule that is important in many processes including the regulation of metabolism and the transporting chemical signals to cells.

4. Vitamin A or retinol is essential for proper functioning of



the organism and in visual clarity of an organism. Once it is taken into the body from diets it is processed or stored. Vitamin A, via 11-cis retinal underlies and is responsible for visual phototransduction. A complex signaling cascade in turn prompting others to relay the signal to centres in the brain where processing of visual perception takes place. For the vision to be sustained and achieved, the all-trans isomerized chromophore must be converted back to the cis form which is accomplished through a series of enzyme-catalyzed reactions collectively known as the retinal or visual cycle.